

STIC Search Report Biotech-Chem Library

STIC Database Tracking Number: 163264

TO: Ben Sackey

Location: REM 5B31/5C18

Art Unit: 1626 August 22, 2005

Case Serial Number: 10/687411

From: P. Sheppard

Location: Remsen Building

Phone: (571) 272-2529

sheppard@uspto.gov

Search Notes	
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24

Scientific and Technical Information Center

SEARCH REQUEST FORM

Requester's Full Name: BEN SACKEY Examiner #: 73489 Date: 81105 Art Unit: 1626 Phone Number: 2-0704 Serial Number: 10/687, 411 Location (Bleg/Room#): 15618 Results Format Preferred (circle): PAPDR DISK ***********************************
To ensure an efficient and quality search, please attach a copy of the cover sheet, claims, and abstract or fill out the following:
Title of Invention: Water Resistent Cutalyst for the production of diayl cubanters (please provide full names): Soloveichik et al.
Earliest Priority Date: 10115103
Search Topic: Please provide a detailed statement of the search topic, and describe as specifically as possible the subject matter to be searched. Include the elected species or structures, keywords, synonyms, acronyms, and registry numbers, and combine with the concept or utility of the invention. Define any terms that may have a special meaning. Give examples or relevant citations, authors, etc., if known.
For Sequence Searches Only Please include all pertinent information (parent, child, divisional, or issued patent numbers) along with the appropriate serial number. method for making a diaryl combanate Compribing Contacting a marking method for making a diaryl comban manoxide and oxygen in the present phenolic compod with comban manoxide and oxygen in the present composition compod with Capitaing palladisin, a co-cetalyst, a base, a holic carbanylettin colonylettin colonylettin addition for increasing the amount of diaryl cubic celebrate and a chemical addition for increasing the amount of diaryl cubic celebrate and a chemical addition for increasing the amount of diaryl cubic celebrate and a chemical addition of additional a
se see the selection of phenolic compols attached.

Sackey 10_687411-History

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		OR TRIGLYME OR TETRAGLYME
		E SOLVENT
L4		SEA ABB=ON PLU=ON SOLVENT OR SOLVENTS
L5	95	SEA ABB=ON PLU=ON NITRILE?/CN
L6	786	SEA ABB=ON PLU=ON AMIDE?/CN
L7		SEA ABB=ON PLU=ON PHENOLIC OR CRESOL OR 4-FLUOROPHENOL?/CN
		OR BISPHENOL A?/CN OR METHYL SALICYLATE?/CN
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L9		SEA ABB=ON PLU=ON L2 OR DIARYL(W) CARBONATE
L10	2183760	SEA ABB=ON PLU=ON L3 OR L4 OR L5 OR L6 OR ACTIVATING (W) SOLVEN
		T OR ETHER? OR SULFONE? OR NITRILE OR AMIDE OR CARBONATE? OR
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L13		SEA ABB=ON PLU=ON REACTANT/RL(L)L10
L14		SEA ABB=ON PLU=ON REACTANT/RL(L)L11
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L17		SEA ABB=ON PLU=ON CARBON MONOXIDE?/CN OR OXYGEN
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		METAL?/CN
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		LUS' ENTERED AT 11:53:08 ON 22 AUG 2005
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L23	1846505	SEA ABB=ON PLU=ON L17 OR CARBON(W)MONOXIDE OR CO OR OXYGEN
		OR O2
L24	904208	SEA ABB=ON PLU=ON L19 OR BASE OR (PHOSPHONIUM OR ?AMMONIUM
		OR LITHIUM OR SODIUM OR POTASSIUM) (3A) HYDROXIDE OR ?AMINE(5A)
		HYDRATE
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L26	2432/1	SEA ABB=ON PLU=ON COPPER?/CN
L27	170458	SEA ABB=ON PLU=ON TITANIUM
		LUS' ENTERED AT 12:01:52 ON 22 AUG 2005
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L29	24	SEA ABB=ON PLU=ON L28 AND PD= <october 14,="" 2003<="" td=""></october>

FILE 'HCAPLUS' ENTERED AT 12:05:46 ON 22 AUG 2005 D STAT QUE

Sackey 10_687411-History

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L30 '	38	SEA	ABB=ON	PLU=ON	L13	AND	L14	AND	L23	AND	L22	AND	L24	AND
		L25												
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FILE HOME

FILE REGISTRY

Property values tagged with IC are from the ZIC/VINITI data file provided by InfoChem.

STRUCTURE FILE UPDATES: 19 AUG 2005 HIGHEST RN 861198-35-8 DICTIONARY FILE UPDATES: 19 AUG 2005 HIGHEST RN 861198-35-8

New CAS Information Use Policies, enter HELP USAGETERMS for details.

TSCA INFORMATION NOW CURRENT THROUGH JANUARY 18, 2005

Please note that search-term pricing does apply when conducting SmartSELECT searches.

Structure search iteration limits have been increased. See HELP SLIMITS for details.

Experimental and calculated property data are now available. For more information enter HELP PROP at an arrow prompt in the file or refer to the file summary sheet on the web at: http://www.cas.org/ONLINE/DBSS/registryss.html

FILE HCAPLUS

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FILE COVERS 1907 - 22 Aug 2005 VOL 143 ISS 9 FILE LAST UPDATED: 21 Aug 2005 (20050821/ED)

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Sackey 10_687411-History

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FILE COVERS 1907 - 22 Aug 2005 VOL 143 ISS 9 FILE LAST UPDATED: 21 Aug 2005 (20050821/ED)

New CAS Information Use Policies, enter HELP USAGETERMS for details.

This file contains CAS Registry Numbers for easy and accurate substance identification.

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L1 STR

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NODE ATTRIBUTES: DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES: RING(S) ARE ISOLATED OR EMBEDDED NUMBER OF NODES IS 6

STEREO ATTRIBUTES: NONE

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L4	1255	SEA FILE=REGISTRY ABB=ON PLU=ON SOLVENT OR SOLVENTS
L5	95	SEA FILE=REGISTRY ABB=ON PLU=ON NITRILE?/CN
L6	786	SEA FILE=REGISTRY ABB=ON PLU=ON AMIDE?/CN
L7	16418	SEA FILE=REGISTRY ABB=ON PLU=ON PHENOLIC OR CRESOL OR
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		CO OR OXYGEN OR O2													
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L29 ANSWER 1 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

2003:154410 HCAPLUS

DOCUMENT NUMBER:

138:187781

TITLE:

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Preparation of 3-phenoxy-4-pyridazinol derivatives as

herbicides

INVENTOR(S):

Tsukamoto, Yoshihisa; Komai, Hiroyuki; Kadotani,

Junji; Koi, Kiyoshi; Mio, Shigeru; Takeshiba, Hideo

PATENT ASSIGNEE(S):

Sankyo Company, Limited, Japan

SOURCE:

PCT Int. Appl., 560 pp. CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

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PATENT 1	NO.			KIN	D 1	DATE		i	APPL:	ICAT:	ION I	NO.	•	DATE			
WO 2003	0162	86		A1	;	2003	0227	1	WO 2	002-	JP82	78		20	0020	314 <	
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	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	ES,	FI,	GB,	GD,	GE,	GH,	
	GM,	HR,	ΗU,	ID,	IL,	IN,	ıs,	JP,	KE,	KG,	ΚP,	KR,	KZ,	LC,	LK,	LR,	
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	PL,	PT,	RO,	RU,	SD,	SE,	SG,	SI,	SK,	SL,	ТJ,	TM,	TN,	TR,	TT,	TZ,	
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RW:	GH,	GM,	KΕ,	LS,	MW,	ΜŻ,	SD,	SL,	SZ,	TZ,	ŪĠ,	ZM,	ZW,	ΑT,	BE,	BG,	

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     US 2005037925
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PRIORITY APPLN. INFO.:
                                              JP 2001-248014
                                                                  Α
                                                                      20010817
                                              JP 2002-82219
                                                                  Α
                                                                      20020325
                                              WO 2002-JP8278
                                                                      20020814
OTHER SOURCE(S):
                          MARPAT 138:187781
GΙ
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The title compds. I [wherein R1 = H, halo, halo(alkyl), cycloalkyl, AΒ alkenyl, CN, alkyl-CO, dialkylcarbamoyl, alkoxy, (un) substituted Ph, 5-6 membered heterocyclyl(oxy), or PhO; R2 = H, halo, (alkoxy)alkyl, alkoxy-CO, trialkylsilyl, (un)substituted PhCO, PhO, or PhS; R3-R7 = independently H, halo, alkynyl, bicycloalkyl, CN, CHO, alkyl-CO, CO2H, alkoxy-CO, (dialkyl)carbamoyl, NO2, OH, (halo)alkoxy, alkoxyalkoxy, alkylthio, alkyl-SO, alkyl-SO2, trialkylsilyl, (un) substituted alkyl, alkenyl, cycloalkyl, PhCO, Ph, 3-6 membered heterocyclyl, amino, PhO, 5-6 membered heterocyclyloxy, or PhSO3; or R3-R7 = neighboring two of them form (un) substituted 3-6 membered cyclohydrocarbyl with the carbon atoms attached; m and n = independently 0 or 1] and salts or ester derivs. thereof are prepared For example, 3,6-dichloropyridazine was coupled with 2-methylphenol in the presence of K2CO3 to give 6-chloro-3-(2-methylphenoxy)pyridazine (57%). pyridazine obtained was treated with POC13 and C12 to produce 4,6-dichloro-3-(2-methylphenoxy)pyridazine (42%). The above compound was hydrolyzed by aqueous NaOH in 1,4-dioxane in the presence of Bu4NCl to afford 6-chloro-3-(2-methylphenoxy)-4-pyridazinol (II) (37%). I showed herbicidal activity, and are useful as herbicides. Formulations containing I as an active ingredient were also described.

IT 499229-28-6P

RN

RL: AGR (Agricultural use); BSU (Biological study, unclassified); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation); USES (Uses)

(herbicide; preparation of phenoxypyridazinol derivs. as herbicides) 499229-28-6 HCAPLUS

Carbonic acid, 6-chloro-3-(2-cyclopropyl-6-methylphenoxy)-4-pyridazinyl CN phenyl ester (9CI) (CA INDEX NAME)

IT 79-22-1, Methyl chlorocarbonate 95-48-7, 2-Methylphenol, reactions 100-39-0, Benzyl bromide 106-96-7, Propargyl bromide 107-30-2, Chloromethoxymethane 108-24-7, Acetic anhydride 135-02-4, 2-Methoxybenzaldehyde 150-19-6, 3-Methoxyphenol 506-68-3, Bromocyanide 920-39-8, Isopropylmagnesium bromide 1195-09-1, 2-Methoxy-5-methylphenol 1310-73-2, Sodium hydroxide, reactions 1779-49-3, Methyltriphenylphosphonium bromide 2219-82-1, 2-tert-Butyl-6-methylphenol 3970-21-6, 2-Methoxyethoxymethyl chloride 7789-59-5, Phosphoric tribromide RL: RCT (Reactant); RACT (Reactant or reagent) (preparation of phenoxypyridazinol derivs. as herbicides) RN79-22-1 HCAPLUS CN Carbonochloridic acid, methyl ester (9CI) (CA INDEX NAME)

RN 95-48-7 HCAPLUS CN Phenol, 2-methyl- (9CI) (CA INDEX NAME)

RN 100-39-0 HCAPLUS CN Benzene, (bromomethyl) - (9CI) (CA INDEX NAME)

Ph-CH2-Br

RN 106-96-7 HCAPLUS CN 1-Propyne, 3-bromo- (9CI) (CA INDEX NAME)

Br-CH2-C≡CH

RN 107-30-2 HCAPLUS CN Methane, chloromethoxy- (9CI) (CA INDEX NAME) C1-CH2-O-CH3

RN 108-24-7 HCAPLUS

CN Acetic acid, anhydride (9CI) (CA INDEX NAME)

Ac-O-Ac

RN 135-02-4 HCAPLUS

CN Benzaldehyde, 2-methoxy- (9CI) (CA INDEX NAME)

RN 150-19-6 HCAPLUS

CN Phenol, 3-methoxy- (9CI) (CA INDEX NAME)

RN 506-68-3 HCAPLUS

CN Cyanogen bromide ((CN)Br) (9CI) (CA INDEX NAME)

Br-C = N

RN 920-39-8 HCAPLUS

CN Magnesium, bromo(1-methylethyl) - (9CI) (CA INDEX NAME)

RN 1195-09-1 HCAPLUS

CN Phenol, 2-methoxy-5-methyl- (9CI) (CA INDEX NAME)

RN 1310-73-2 HCAPLUS

CN Sodium hydroxide (Na(OH)) (9CI) (CA INDEX NAME)

Na-OH

RN 1779-49-3 HCAPLUS

CN Phosphonium, methyltriphenyl-, bromide (8CI, 9CI) (CA INDEX NAME)

• Br-

RN 2219-82-1 HCAPLUS

CN Phenol, 2-(1,1-dimethylethyl)-6-methyl- (9CI) (CA INDEX NAME)

RN 3970-21-6 HCAPLUS

CN Ethane, 1-(chloromethoxy)-2-methoxy- (7CI, 8CI, 9CI) (CA INDEX NAME)

 $MeO-CH_2-CH_2-O-CH_2C1$

RN 7789-59-5 HCAPLUS

CN Phosphoric tribromide (9CI) (CA INDEX NAME)

REFERENCE COUNT:

2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 2 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

2003:77802 HCAPLUS

DOCUMENT NUMBER:

138:124222

TITLE:

Process and catalyst systems for the carbonylation manufacture of diaryl carbonates from phenols and

carbon monoxide and dioxide

INVENTOR (S):

Reisinger, Claus-Peter; Hansen, Sven Michael; Fischer,

PATENT ASSIGNEE(S):

Bayer A.-G., Germany; Bayer Materialscience A.-G.

SOURCE: Eur. Pat. Appl., 9 pp.

CODEN: EPXXDW

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT:

1

PATENT INFORMATION:

PATENT NO.	KIND DATE	APPLICATION NO.	DATE
EP 1279659	A2. 20030129	EP 2002-15584	20020715 <
EP 1279659	A3 20040303		
R: AT, BE, CH,	DE, DK, ES, FR,	GB, GR, IT, LI, LU, NL,	SE, MC, PT,
IE, SI, LT,	LV, FI, RO, MK,	CY, AL, TR, BG, CZ, EE,	SK
DE 10136856	A1 20030213	DE 2001-10136856	20010727 <
SG 103877	A1 20040526	SG 2002-4323	20020712
JP 2003096027	A2 20030403	JP 2002-211168	20020719 <
US 2003036663	A1 20030220	US 2002-200667	20020722 <
US 6852872	B2 20050208		
BR 2002002955	A 20030603	BR 2002-2955	20020725 <
CN 1400204	A 20030305	CN 2002-127060	20020726 <
PRIORITY APPLN. INFO.:		DE 2001-10136856	A 20010727
OTHER SOURCE(S):	MARPAT 138:1242:	22	

A process and for the carbonylation manufacture of diaryl carbonates (e.g., di-Ph carbonate) from phenols (e.g., phenol) and carbon monoxide and dioxide is conducted in the presence of a catalyst system comprising a Group VIIIB metal salt (e.g., palladium dibromide) where there are at least two metal salts (e.g., manganese trisacetylacetonate) and a base (e.g., tetrabutylammonium bromide).

1643-19-2, Tetrabutylammonium bromide 13444-94-5, IT

Palladium dibromide

RL: CAT (Catalyst use); USES (Uses)

(catalysts for the carbonylation manufacture of diaryl carbonates from phenols and carbon monoxide and dioxide)

RN1643-19-2 HCAPLUS

CN 1-Butanaminium, N,N,N-tributyl-, bromide (9CI) (CA INDEX NAME)

Br⁻

RN13444-94-5 HCAPLUS

Palladium bromide (PdBr2) (7CI, 8CI, 9CI) (CA INDEX NAME)

Br-Pd-Br

IT 108-95-2, Phenol, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (catalysts for the carbonylation manufacture of diaryl carbonates from phenols and carbon monoxide and dioxide) 108-95-2--HCAPLUS--RN-Phenol (8CI, 9CI) (CA INDEX NAME) CNOH 102-09-0P, Diphenyl carbonate IT RL: IMF (Industrial manufacture); PREP (Preparation) (process and catalyst systems for the carbonylation manufacture of diaryl carbonates from phenols and carbon monoxide and dioxide) 102-09-0 HCAPLUS RNCN Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME) PhO-C-OPh 630-08-0, Carbon monoxide, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (process and catalyst systems for the carbonylation manufacture of diaryl carbonates from phenols and carbon monoxide and dioxide) 630-08-0 HCAPLUS RN Carbon monoxide (8CI, 9CI) (CA INDEX NAME) CN L29 ANSWER 3 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN ACCESSION NUMBER: 2002:907214 HCAPLUS DOCUMENT NUMBER: 137:386324 TITLE: Method and catalysts for producing aromatic carbonate esters from phenols and carbon monoxide Pressman, Eric James; Ofori, John Yaw INVENTOR(S): General Electric Company, USA PATENT ASSIGNEE(S): U.S. Pat. Appl. Publ., 16 pp. SOURCE: CODEN: USXXCO Patent DOCUMENT TYPE: English LANGUAGE: FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

APPLICATION NO.

DATE

KIND

DATE

PATENT NO.

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     US 6800779
                          B2
                                20041005
PRIORITY APPLN. INFO.:
                                            US 2001-822531
                                                                   20010330
     A method for economically producing aromatic carbonates (e.g., di-Ph
     <u>carbonate) from aromatic hydroxy compds. (e.g., phenol) is described which in </u>
     one embodiment comprises the steps of: (i) contacting, at a temperature
     sufficient to keep the mixture molten, at least one aromatic hydroxy compound
     with a catalyst composition comprising the following and any reaction products
     thereof: (A) at least one Group VIII metal or a compound; (B) at least one
     salt; (C) at least one metal co-catalyst; and (D) optionally, at
     least one activating solvent; (ii) optionally heating the mixture at
atmospheric
     pressure to a temperature above that sufficient to keep the mixture molten;
(iii)
     pressurizing the mixture with carbon monoxide; (iv)
     optionally heating the mixture under pressure of carbon
     monoxide to a temperature above that sufficient to keep the mixture
     molten; (v) optionally maintaining the mixture under pressure of
     carbon monoxide for a time period; (vi) introducing
     oxygen to the mixture to a desired concentration of oxygen in
     carbon monoxide; (vii) starting gas flow to the mixture at
     a desired concentration of oxygen and carbon monoxide
     ; (viii) optionally maintaining gas flow for a time period at less than a
     desired ultimate temperature for the mixture; and (ix) optionally heating the
     mixture to a desired ultimate temperature under flow of gases.
     1310-73-2, Sodium hydroxide, uses
IT
     RL: CAT (Catalyst use); USES (Uses)
        (base; catalysts for producing aromatic carbonate esters from
        phenols and carbon monoxide)
     1310-73-2 HCAPLUS
RN
     Sodium hydroxide (Na(OH)) (9CI)
                                      (CA INDEX NAME)
CN
Na-OH
                                ?
     89610-32-2, Hexaethylguanidinium bromide
IT
     RL: CAT (Catalyst use); USES (Uses)
         (catalyst for producing aromatic carbonate esters from phenols and
        carbon monoxide)
RN
     89610-32-2 HCAPLUS
     Ethanaminium, N-[bis(diethylamino)methylene]-N-ethyl-, bromide (9CI) (CA
CN
     INDEX NAME)
      N+Et2
 Et2N-C-NEt2
     Br
```

7647-15-6 HCAPLUS RN

Sodium bromide (NaBr) (9CI) (CA INDEX NAME) CN

Br Na

IT 102-09-0P, Diphenyl carbonate

RL: IMF (Industrial manufacture); PREP (Preparation) (method and catalysts for producing aromatic carbonate esters from phenols and carbon monoxide)

102-09-0 HCAPLUS RN

Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME) CN

Pho-c-oph

IT 108-95-2, Phenol, reactions 630-08-0,

Carbon monoxide, reactions

RL: RCT (Reactant); RACT (Reactant or reagent) (method and catalysts for producing aromatic carbonate esters from phenols and carbon monoxide) -

108-95-2 HCAPLUS RN

CN Phenol (8CI, 9CI) (CA INDEX NAME)

OH

630-08-0 HCAPLUS RN

Carbon monoxide (8CI, 9CI) (CA INDEX NAME) CN

IT7782-44-7, Oxygen, reactions

RL: RGT (Reagent); RACT (Reactant or reagent) (method and catalysts for producing aromatic carbonate esters from phenols and carbon monoxide using)

RN 7782-44-7 HCAPLUS

Oxygen (8CI, 9CI) (CA INDEX NAME) CN

o = o

REFERENCE COUNT: 35 THERE ARE 35 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 4 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER: 2002:900782 HCAPLUS

DOCUMENT NUMBER:

138:4417

TITLE:

Preparation of diaryl carbonates

INVENTOR(S):

Tange, Shinya; Ohashi, Kenji; Nagashima, Ryoichi;

Yoshizato, Hidenobu

PATENT ASSIGNEE(S):

Teijin Ltd., Japan

SOURCE:

Jpn. Kokai Tokkyo Koho, 9 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
JP 2002338525	A2	20021127	JP 2001-144324	20010515 <		
PRIORITY APPLN. INFO.:			JP 2001-144324	20010515		

OTHER SOURCE(S): MARPAT 138:4417 In preparation of R2CO3 [R = (un)substituted C6-15 aryl], useful as materials for aromatic polycarbonates, by oxidative carbonylation of ROH (R = same as above) with CO and O2 in the presence of catalysts and inert substance (A) while removing H2O formed during reaction together with (A), (A) is recovered from byproduct (Y) through a process involving (1) ≥ 1 step to decrease water content and (2) ≥ 1 step to distill (A). This method increases conversion, selectivity, or yield and the recovered (A) can be reused in the reaction. A mixture containing PhOH, THF, Pd(OAc)2, Mn(OAc)2, Bu4N+ Br-, and (Bu4N)4SiWMo11040 was bubbled with CO and O2 at 80° and 0.780 MPa for 5 h to give Ph2CO3 containing 0.2% H2O at selectivity 98.7%. Mixed vapor formed during the reaction was continuously fed to a PhOH trap and mixed vapor passed through the trap was introduced to a condenser to recover co and 02 for reuse. The condensate containing 96% THF and 4% H2O was fed to the bottom of an extraction column, where aqueous NaOH solution was fed from

at 40° and 0.1 MPa to give THF containing 0.4% H2O. The recovered THF was distilled using a batch distillation column to obtain THF containing ≤30 ppm

H2O from the bottom.

IT 1643-19-2, Tetrabutylammonium bromide

RL: CAT (Catalyst use); USES (Uses)

(catalyst; preparation of diaryl carbonates by oxidative carbonylation of phenols under azeotropic removal of H2O)

RN1643-19-2 HCAPLUS

CN1-Butanaminium, N,N,N-tributyl-, bromide (9CI) (CA INDEX NAME)

● Br-

TΥ 102-09-0P, Diphenyl carbonate

RL: IMF (Industrial manufacture); SPN (Synthetic preparation); PREP

(preparation of diaryl carbonates by oxidative

carbonylation of phenols under azeotropic removal of H2O)
RN 102-09-0 HCAPLUS
CN Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

PhO-C-OPh

IT 108-95-2, Phenol, reactions 630-08-0, Carbon monoxide, reactions 7782-44-7,

Oxygen, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(preparation of diaryl carbonates by oxidative carbonylation of phenols under azeotropic removal of H2O)

RN 108-95-2 HCAPLUS

CN Phenol (8CI, 9CI) (CA INDEX NAME)

OH

RN 630-08-0 HCAPLUS

CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

C +

RN 7782-44-7 HCAPLUS

CN Oxygen (8CI, 9CI) (CA INDEX NAME)

o = o

IT 1310-73-2, Sodium hydroxide, uses

RL: NUU (Other use, unclassified); USES (Uses)

(water removal by extraction with solution of; preparation of diaryl carbonates by

oxidative carbonylation of phenols under azeotropic removal of H2O)

RN 1310-73-2 HCAPLUS

CN Sodium hydroxide (Na(OH)) (9CI) (CA INDEX NAME)

Na-OH

L29 ANSWER 5 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

2002:551630 HCAPLUS

DOCUMENT NUMBER:

137:95535

TITLE:

Method of sustaining catalyst activity in the oxidative carbonylation catalytic production of

and the second s

aromatic carbonates Pressman, Eric James

INVENTOR(S):
PATENT ASSIGNEE(S):

SOURCE:

General Electric Company, USA

U.S., 7 pp. CODEN: USXXAM

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. ______ -------------------US 6423863 B1 20020723 US 2001-681940 20010628 <--WO 2003002507 **A**1 20030109 WO 2002-US11797 20020410 <--AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG US 2001-681940 PRIORITY APPLN. INFO.: A 20010628

AB The present invention is directed to a method for sustaining the catalytic activity of a carbonylation catalyst composition, after changes in reactor pressure and temperature, in the catalytic production of aromatic carbonates (e.g.,

di-Ph carbonate).

IT 75-59-2, Tetramethylammonium hydroxide

77-98-5, Tetraethylammonium hydroxide

1310-58-3, Potassium hydroxide, reactions

1310-65-2, Lithium hydroxide 1310-73-2

, Sodium hydroxide, reactions 32680-30-1,

Methyltributylammonium hydroxide

RL: RGT (Reagent); RACT (Reactant or reagent)

(base; method of sustaining catalyst activity in the

oxidative carbonylation catalytic production of aromatic carbonates)

RN 75-59-2 HCAPLUS

CN Methanaminium, N,N,N-trimethyl-, hydroxide (9CI) (CA INDEX NAME)

OH-

RN 77-98-5 HCAPLUS

CN Ethanaminium, N,N,N-triethyl-, hydroxide (9CI) (CA INDEX NAME)

● OH-

RN 1310-58-3 HCAPLUS

CN Potassium hydroxide (K(OH)) (9CI) (CA INDEX NAME)

K-OH

RN 1310-65-2 HCAPLUS

CN Lithium hydroxide (Li(OH)) (9CI) (CA INDEX NAME)

Li-OH

RN 1310-73-2 HCAPLUS

CN Sodium hydroxide (Na(OH)) (9CI) (CA INDEX NAME)

Na-OH

RN 32680-30-1 HCAPLUS

CN 1-Butanaminium, N, N-dibutyl-N-methyl-, hydroxide (9CI) (CA INDEX NAME)

● OH-

IT 64-20-0, Tetramethylammonium bromide 71-91-0,

Tetraethylammonium bromide 7550-35-8, Lithium bromide

7647-15-6, Sodium bromide, uses 89610-32-2,

Hexaethylguanidinium bromide

RL: CAT (Catalyst use); USES (Uses)

(method of sustaining catalyst activity in the oxidative carbonylation catalytic production of aromatic carbonates)

RN 64-20-0 HCAPLUS

CN Methanaminium, N,N,N-trimethyl-, bromide (9CI) (CA INDEX NAME)

• Br

RN 71-91-0 HCAPLUS CN Ethanaminium, N,N,N-triethyl-, bromide (9CI) (CA INDEX NAME)

● Br⁻

RN 7550-35-8 HCAPLUS CN Lithium bromide (LiBr) (9CI) (CA INDEX NAME)

Br-Li

RN 7647-15-6 HCAPLUS CN Sodium bromide (NaBr) (9CI) (CA INDEX NAME)

Br-Na

RN 89610-32-2 HCAPLUS
CN Ethanaminium, N-[bis(diethylamino)methylene]-N-ethyl-, bromide (9CI) (CA INDEX NAME)

● Br-

IT 102-09-0P, Diphenyl carbonate
RL: IMF (Industrial manufacture); PREP (Preparation)
(method of sustaining catalyst activity in the oxidative carbonylation

catalytic production of aromatic carbonates)

RN 102-09-0 HCAPLUS

CN Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

IT 108-95-2, Phenol, reactions 630-08-0,

Carbon monoxide, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(method of sustaining catalyst activity in the oxidative carbonylation catalytic production of aromatic carbonates)

RN 108-95-2 HCAPLUS

CN Phenol (8CI, 9CI) (CA INDEX NAME)

OH

RN 630-08-0 HCAPLUS

CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

C +

REFERENCE COUNT: 1 THERE ARE 1 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 6 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

2002:466747 HCAPLUS

DOCUMENT NUMBER:

INVENTOR(S):

137:33683

TITLE:

An improved process for removing water from oxidative

carbonylation in production of diaryl carbonates Ofori, John Yaw; Pressman, Eric James; Shalyaev, Kirill Vladimirovich; Williams, Eric Douglas;

Battista, Richard Anthony

PATENT ASSIGNEE(S):

General Electric Company, USA

SOURCE:

U.S. Pat. Appl. Publ., 20 pp., Cont.-in-part of U.S.

Ser. No. 736,885.

CODEN: USXXCO

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2002077497	A1	20020620	US 2001-961747	20010924 <
US 6420589	B2	20020716		
WO 2002048088	A2	20020620	WO 2001-US47205	20011113 <
WO 2002048088	A3	20021219		

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WO 2002048088
                          B1
                                20030130
            AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
             CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR,
            HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT,
             LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT,
             RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ,
             VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY,
            DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF,
             BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
                                20020624
                                          AU 2002-26018
    AU 2002026018
                         Α5
                                                                   20011113 <--
    DE 10197052
                          Т
                                20031023
                                            DE 2001-10197052
                                                                   20011113
     JP 2004526678
                         T2
                                20040902
                                            JP 2002-549624
                                                                   20011113
     US 2002111507
                         A1
                                20020815
                                            US 2002-121102
                                                                   20020411 <--
    US 6472551
                         B2
                                20021029
PRIORITY APPLN. INFO.:
                                            US 2000-736885
                                                                A2 20001214
                                            US 2001-961747
                                                                A 20010924
                                            WO 2001-US47205
                                                                W 20011113
     The process comprises: (1) contacting at least one aromatic hydroxy compound
AB
     with carbon monoxide and oxygen in the
    presence of a catalyst composition (I), (2) removing a liquid stream (L) from
the
     reaction vessel, (3) transferring L to a flash vessel to remove the
     majority of water under reduced pressure, and (4) returning at least a
    portion of a dried L back to the reaction vessel, wherein at least a
    portion of diaryl carbonate is recovered from L either before or after
     water removal and I contains: (A) at least one metal having an atomic number
     ≥44 from Group 8, 9, or 10, (B) at least one alkali metal salt, (C)
     at least one metal cocatalyst, (D) at least one activating organic solvent,
    and (E) optionally one base.
     1310-73-2, Sodium hydroxide, uses
     7647-15-6, Sodium bromide, uses 13444-94-5, Palladium
     bromide
     RL: CAT (Catalyst use); USES (Uses)
        (in production of di-Ph carbonate by oxidative carbonylation)
RN
     1310-73-2 HCAPLUS
     Sodium hydroxide (Na(OH)) (9CI) (CA INDEX NAME)
CN
Na-OH
     7647-15-6 HCAPLUS
RN
     Sodium bromide (NaBr) (9CI) (CA INDEX NAME)
Br Na
RN
     13444-94-5 HCAPLUS
     Palladium bromide (PdBr2) (7CI, 8CI, 9CI) (CA INDEX NAME)
Br-Pd-Br
     108-95-2, Phenol, reactions 630-08-0,
IT
     Carbon monoxide, reactions 7782-44-7,
     Oxygen, reactions
```

RL: RCT (Reactant); RACT (Reactant or reagent)

(in production of di-Ph carbonate by oxidative carbonylation)

RN 108-95-2 HCAPLUS

CN Phenol (8CI, 9CI) (CA INDEX NAME)

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OH
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RN 630-08-0 HCAPLUS

CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

C +

RN 7782-44-7 HCAPLUS

CN Oxygen (8CI, 9CI) (CA INDEX NAME)

o = 0

IT 102-09-0P, Diphenyl carbonate

RL: IMF (Industrial manufacture); PREP (Preparation)

(production of di-Ph carbonate by oxidative carbonylation)

RN 102-09-0 HCAPLUS

CN Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

Pho-C-OPh

L29 ANSWER 7 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

2002:466746 HCAPLUS

DOCUMENT NUMBER:

137:33682

TITLE:

A method for production of diaryl carbonates by oxidative carbonylation with removal of undesired

water during the reaction

INVENTOR(S):

Ofori, John Yaw; Pressman, Eric James; Shalyaev, Kirill Vladimirovich; Williams, Eric Douglas;

Battista, Richard Anthony

PATENT ASSIGNEE(S):

General Electric Company, USA

SOURCE:

U.S. Pat. Appl. Publ., 16 pp., Cont.-in-part of U.S.

Ser. No. 736,751, abandoned.

CODEN: USXXCO

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

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US 2001-961745
                                20020620
                                                                    20010924 <--
    US 2002077496
                          A1
                                20030218
    US 6521777
                          B2
                                            WO 2001-US43496
    WO 2002048087
                          A2
                                20020620
                                                                    20011114 <--
                                20030213
    WO 2002048087
                          Α3
            AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
             CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
             GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
             LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH,
             PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA,
             UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY,
             DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF,
             BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG
    AU 2002016685
                                20020624
                                            AU 2002-16685
                          A5
                                                                    20011114 <--
    DE 10197050
                          \mathbf{T}
                                20040429
                                             DE 2001-10197050
                                                                    20011114
                          T2
                                             JP 2002-549623
     JP 2004521094
                                20040715
                                                                    20011114
                                             US 2000-736751
PRIORITY APPLN. INFO.:
                                                                 B2 20001214
                                             US 2001-961745
                                                                 Α
                                                                    20010924
                                             WO 2001-US43496
                                                                 W
                                                                    20011114
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AB The method comprises: (1) contacting at least one aromatic hydroxy compound with carbon monoxide and oxygen in the presence of a catalyst composition (I), (2) removing a liquid stream (L) from

the

agitating reaction mixture in a vessel, and transferring L to a first disengagement vessel without agitating, (3) transferring L then to a flash vessel to remove the majority of water under reduced pressure, (4) returning at least a portion of a dried L back to the reaction vessel, wherein at least a portion of diaryl carbonate is recovered from L either before or after water removal and I contains: (A) at least one metal having an atomic number ≥44 from Group 8, 9, or 10, (B) at least one guanidinium salt or onium salt, (C) at least one metal cocatalyst, and (D) at least one base.

71-91-0, Tetraethylammonium bromide 1310-73-2, Sodium hydroxide, uses 13444-94-5, Palladium bromide

RL: CAT (Catalyst use); USES (Uses)

(in production of di-Ph carbonate by oxidative carbonylation)

RN 71-91-0 HCAPLUS

CN Ethanaminium, N,N,N-triethyl-, bromide (9CI) (CA INDEX NAME)

● Br-

RN 1310-73-2 HCAPLUS

CN Sodium hydroxide (Na(OH)) (9CI) (CA INDEX NAME)

Na-OH

RN 13444-94-5 HCAPLUS

CN Palladium bromide (PdBr2) (7CI, 8CI, 9CI) (CA INDEX NAME)

Br-Pd-Br

ОН

RN 630-08-0 HCAPLUS CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

C +

IT 102-09-0P, Diphenyl carbonate
RL: IMF (Industrial manufacture); PREP (Preparation)

(production of di-Ph carbonate by oxidative carbonylation)

RN 102-09-0 HCAPLUS CN Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

O || PhO- C- OPh

INVENTOR(S):

L29 ANSWER 8 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER: 2002:466745 HCAPLUS

DOCUMENT NUMBER: 137:33681

TITLE: A method for removal of undesired water from oxidative

carbonylation in production of diaryl carbonates Ofori, John Yaw; Pressman, Eric James; Shalyaev,

Kirill Vladimirovich; Williams, Eric Douglas;

Battista, Richard Anthony

PATENT ASSIGNEE(S): US

SOURCE: U.S. Pat. Appl. Publ., 11 pp.

CODEN: USXXCO

DOCUMENT TYPE: Patent LANGUAGE: English

LANGUAGE: Englis FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

A1 20020620 US 2000-736872 20001214 <--US 2002077495 PRIORITY APPLN. INFO.: US 2000-736872 20001214 The method comprises: (1) contacting at least one aromatic hydroxy compound with carbon monoxide and oxygen in the presence of a catalyst composition (I), (2) removing a liquid stream (L) from the_ reaction vessel, (3) transferring L to a flash vessel to remove the majority of water under reduced pressure, and (4) returning at least a portion of a dried L back to the reaction vessel, wherein at least a portion of diaryl carbonate is recovered from L either before or after water removal and I contains: (A) at least one metal having an atomic number ≥44 from Group 8, 9, or 10, (B) at least one guanidinium salt or onium salt, (C) at least one metal cocatalyst, and (D) at least one base. 71-91-0, Tetraethylammonium bromide 1310-73-2, IT Sodium hydroxide, uses 13444-94-5, Palladium bromide RL: CAT (Catalyst use); USES (Uses) (in production of di-Ph carbonate by oxidative carbonylation) RN71-91-0 HCAPLUS Ethanaminium, N,N,N-triethyl-, bromide (9CI) (CA INDEX NAME) CNΕt Et-N+Et Εt ● Br-1310-73-2 HCAPLUS RNSodium hydroxide (Na(OH)) (9CI) (CA INDEX NAME) CN Na-ОН RN 13444-94-5 HCAPLUS Palladium bromide (PdBr2) (7CI, 8CI, 9CI) (CA INDEX NAME) CN Br-Pd-Br 108-95-2, Phenol, reactions 630-08-0, TT

(in production of di-Ph carbonate by oxidative carbonylation)

Carbon monoxide, reactions.

108-95-2 HCAPLUS

Phenol (8CI, 9CI)

RN

CN

RL: RCT (Reactant); RACT (Reactant or reagent)

(CA INDEX NAME)

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OH
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RN 630-08-0 HCAPLUS

CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

IT 102-09-0P, Diphenyl carbonate

RL: IMF (Industrial manufacture); PREP (Preparation)

(production of di-Ph carbonate by oxidative carbonylation)

RN 102-09-0 HCAPLUS

CN Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

PhO-C-OPh

L29 ANSWER 9 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

2002:428851 HCAPLUS

DOCUMENT NUMBER:

137:7789

TITLE:

Method and catalyst system for producing aromatic carbonates by carbonylation of aromatic hydroxy

compounds

INVENTOR(S):

Shalyaev, Kirill Vladimirovich; Soloveichik, Grigorii

Lev; Johnson, Bruce Fletcher

PATENT ASSIGNEE(S):

General Electric Company, .USA

SOURCE:

PCT Int. Appl., 24 pp. CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.		KIND DATE				APPLICATION NO.						DATE				
THE DIT NO.			1/11/1					AL I 1	ICAI.	IOI I			DATE			
				-									-			
WO 2002044122			A2	:	20020606 WO			WO 2001-US50668				20011019 <				
WO 20020443	WO 2002044122		A3	:	20020906											
W: AE	AG,	ΑL,	AM,	AT,	AU,	ΑZ,	BA,	BB,	BG,	BR,	BY,	BZ,	CA,	CH,	CN,	
CO	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	ES,	FI,	GB,	GD,	GE,	GH,	
GM,	HR,	ΗU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KP,	KR,	ΚZ,	LC,	LK,	LR,	
LS	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NO,	NZ,	PH,	PL,	
PT	RO,	RU,	SD,	SE,	SG,	SI,	SK,	SL,	TJ,	TM,	TR,	TT,	TZ,	UA,	UG,	
UZ,	VN,	YU,	ZA,	ZW,	AM,	ΑZ,	BY,	KG,	KZ,	MD,	RU,	TJ,	TM			
RW: GH	GM,	KE,	LS,	MW,	MZ,	SD,	SL,	SZ,	TZ,	UG,	ZW,	ΑT,	BE,	CH,	CY,	
DE,	DK,	ES,	FI,	FR,	GB,	GR,	ΙE,	IT,	LU,	MC,	NL,	PT,	SE,	TR,	BF,	
BJ	CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	ΝE,	SN,	TD,	TG		
US 20020992	35		Ą1	:	2002	0725	1	US 2	000	7282	24		20	0001	130 <-	
US 6566295			B2	•	2003	0520										

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A5
                                20020611
                                            AU 2002-34135
                                                                    20011019 <--
     AU 2002034135
                                            DE 2001-10196977
                                                                    20011019
    DE 10196977
                          Т
                                20040422
                                            JP 2002-546492
                          T2
                                20041125
                                                                    20011019
     JP 2004535267
                          A1
                                20021205
                                            US 2002-151334
                                                                    20020520 <--
     US 2002183539
                          B2
                                20030128
     US 6512134
                                                                 A 20001130
                                            US 2000-728224
PRIORITY APPLN _INFO .:
                                            . WO 2001-US50668
     The method comprises by reacting ≥1 aromatic hydroxy compound (e.g.,
AB
     phenol) with oxygen and carbon monoxide in .
     the presence of a carbonylation catalyst system containing ≥1 Group 8,
     9 or 10 metal source (e.g., palladium acetylacetonate), ≥1 bromide
     composition (e.g., sodium bromide), ≥1 activating organic solvent (e.g.,
     tetraglyme), a combination of inorg. cocatalysts comprising ≥1
     titanium source (e.g., titanium oxide acetylacetonate) and ≥1
     copper source (e.g., copper acetylacetonate) and ≥1 base
     (e.g., sodium hydroxide) to form an aromatic carbonate
     (e.g., aromatic carbonate).
     1310-73-2, Sodium hydroxide, uses
IT
     7647-15-6, Sodium bromide, uses
     RL: CAT (Catalyst use); USES (Uses)
        (method and catalyst system for producing aromatic carbonates by
        carbonylation of aromatic hydroxy compds.)
     1310-73-2 HCAPLUS
RN
     Sodium hydroxide (Na(OH)) (9CI) (CA INDEX NAME)
CN
Na-OH
RN
     7647-15-6 HCAPLUS
     Sodium bromide (NaBr) (9CI) (CA INDEX NAME)
CN
Br-Na
     102-09-0P, Diphenyl carbonate
IT
     RL: IMF (Industrial manufacture); PREP (Preparation)
         (method and catalyst system for producing aromatic carbonates by
        carbonylation of aromatic hydroxy 'compds.)
RN
     102-09-0 HCAPLUS
     Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)
CN
Pho-C-OPh
     108-95-2, Phenol, reactions 630-08-0,
IT
     Carbon monoxide, reactions 7782-44-7,
     Oxygen, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
         (method and catalyst system for producing aromatic carbonates by
        carbonylation of aromatic hydroxy compds.)
RN
      108-95-2 HCAPLUS
      Phenol (8CI, 9CI) (CA INDEX NAME)
CN
```



RN 630-08-0 HCAPLUS

CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

C -∭ +

RN 7782-44-7 HCAPLUS

CN Oxygen (8CI, 9CI) (CA INDEX NAME)

0==0

L29 ANSWER 10 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER: 2002:129105 HCAPLUS

DOCUMENT NUMBER: 136:183616

TITLE: Reactivation of catalysts and preparation of aromatic

carbonates with the reactivated catalysts

INVENTOR(S): Yoshisato, Akinobu; Muramoto, Masaharu; Ban, Tetsuo

PATENT ASSIGNEE(S): Teijin Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2002053526	A2	20020219	JP 2000-238250	20000807 <
PRIORITY APPLN. INFO.:			JP 2000-238250	20000807

OTHER SOURCE(S): CASREACT 136:183616

AB Solid catalysts comprising Pt-group metals, their compds., or their complexes supported on carriers, which have been used in preparation of aromatic

carbonates by treatment of aromatic hydroxy compds. with CO and O in the presence of quaternary ammonium salts or phosphonium salts and optional bases, are reactivated by treating with the aromatic hydroxy compds. (and their mixts. with organic solvents). Thus, PhOH was treated with Bu4NBr, Mn(II) acetylacetonate, and Pd supported on perovskite-type La0.2Pb0.8ZrO3 under CO and O at 80° and 10 bar for 3 h to give 15.8% di-Ph carbonate. The catalyst was recovered, washed with PhOH, and reused to show almost the same activity as the fresh catalyst.

IT 1643-19-2, Tetrabutylammonium bromide

RL: CAT (Catalyst use); USES (Uses)

(reactivation of catalysts in preparation of aromatic carbonates)

RN 1643-19-2 HCAPLUS

CN 1-Butanaminium, N,N,N-tributyl-, bromide (9CI) (CA INDEX NAME)

Br⁻

RN 102-09-0 HCAPLUS CN Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

IT 630-08-0, Carbon monoxide, reactions
7782-44-7, Oxygen, reactions
RL: RCT (Reactant); RACT (Reactant or reagent)
(reactivation of catalysts in preparation of aromatic carbonates)
RN 630-08-0 HCAPLUS
CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

C-|| +

RN 7782-44-7 HCAPLUS CN Oxygen (8CI, 9CI) (CA INDEX NAME)

o = o

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L29 ANSWER 11 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN
ACCESSION NUMBER:
                         2001:111542 HCAPLUS
DOCUMENT NUMBER:
                         134:149297
TITLE:
                         Carbonylation method and catalyst system for producing
                         -aromatic-carbonates-from-hydroxyaromatic-compounds.---
                         oxygen and carbon monoxide
INVENTOR (S):
                         Patel, Ben Purushotam; Soloveichik, Grigorii Lev;
                         Whisenhunt, Donald Wayne, Jr.; Shalyaev, Kirill
                         Vladimirovich
PATENT ASSIGNEE(S):
                         General Electric Company, USA
SOURCE:
                         U.S., 7 pp.
                         CODEN: USXXAM
DOCUMENT TYPE:
                         Patent
                         English
LANGUAGE:
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
     PATENT NO.
                         KIND
                                DATE
                                            APPLICATION NO.
                                                                   DATE
                         ----
                                _____
                                            ______
     US 6187942
                          _{\rm B1}
                                20010213
                                            US 2000-517000
                                                                   20000301 <--
                                            US 2000-729123
     US 2001031888
                          A1
                                20011018
                                                                   20001204 <--
     US 6355824
                          B2
                                20020312
     WO 2001064617
                          A1
                                20010907
                                            WO 2001-US839
                                                                   20010111 <--
            AL; AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE,
             DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP,
             KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN,
             MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM,
             TR, TT, UA, UG, UZ, VN, YU, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY,
             DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF,
             BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                                20021211
                                            EP 2001-955099
                                                                    20010111 <--
     EP 1263710
                          A1
            AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
                                            JP 2001-563461
     JP 2003525262
                                20030826
                                                                    20010111 <--
                          T2
PRIORITY APPLN. INFO.:
                                            US 2000-517000
                                                                 A3 20000301
                                            WO 2001-US839
                                                                W 20010111
     Aromatic hydroxy compds. (e.g., phenol) are carbonylated into diaryl
AΒ
     carbonates (e.g., di-Ph carbonate) by contacting them with oxygen
     and carbon monoxide in the presence of a carbonylation
     catalyst system comprising an iron compound (e.g., ferrous acetate) as the
     primary catalyst component, and an inorg. cocatalyst (e.g.,
     tetraethylammonium chloride). This process does not use costly
     platinum-group metal compound catalysts; a process flow diagram is
     presented.
     71-91-0, Tetraethylammonium bromide 1310-73-2,
IT
     Sodium hydroxide, uses
     RL: CAT (Catalyst use); USES (Uses)
        (carbonylation cocatalysts for producing aromatic carbonates from
        hydroxyarom. compds., oxygen and carbon
        monoxide)
     71-91-0 HCAPLUS
RN
     Ethanaminium, N,N,N-triethyl-, bromide (9CI)
                                                   (CA INDEX NAME)
CN
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Εt
Et-N+Et
 Br - .
RN
     1310-73-2 HCAPLUS
CN
     Sodium hydroxide (Na(OH)) (9CI)
                                      (CA INDEX NAME)
Na-OH
     10031-26-2, Ferric bromide
IT
     RL: CAT (Catalyst use); USES (Uses)
        (carbonylation method and catalyst system for producing aromatic
        carbonates from hydroxyarom. compds. and oxygen and
        carbon monoxide)
     10031-26-2 HCAPLUS
RN
     Iron bromide (FeBr3) (8CI, 9CI)
CN
                                       (CA INDEX NAME)
   Br
Br-Fe-Br
     102-09-0P, Diphenyl carbonate
IT
     RL: IMF (Industrial manufacture); SPN (Synthetic preparation); PREP
     (Preparation)
        (carbonylation method and catalyst system for producing aromatic
        carbonates from hydroxyarom. compds. and oxygen and
        carbon monoxide)
RN
     102-09-0 HCAPLUS
     Carbonic acid, diphenyl ester (6CI, 8CI, 9CI)
CN
                                                    (CA INDEX NAME)
Pho-C-OPh
IT
     108-95-2, Phenol, reactions 630-08-0,
     Carbon monoxide, reactions 7782-44-7,
     Oxygen, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (carbonylation method and catalyst system for producing aromatic
        carbonates from hydroxyarom. compds. and oxygen and
```

carbon monoxide)

108-95-2 HCAPLUS Phenol (8CI, 9CI)

RN

CN

(CA INDEX NAME)



RN 630-08-0 HCAPLUS

CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

C -

RN 7782-44-7 HCAPLUS

CN Oxygen (8CI, 9CI) (CA INDEX NAME)

o = o

REFERENCE COUNT:

THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 12 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

2001:91544 HCAPLUS

DOCUMENT NUMBER:

134:149285

TITLE:

Method and catalyst system for producing aromatic

carbonates

INVENTOR(S):

Patel, Ben Purushotam; Soloveichik, Grigorii Lev;

Whisenhunt, Donald Wayne, Jr.; Shalyaev, Kirill

Vladimirovich

PATENT ASSIGNEE(S):

General Electric Company, USA

SOURCE:

U.S., 7 pp. CODEN: USXXAM

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND DATE	APPLICATION NO.	DATE
US 6184409	B1 20010206	US 2000-516746	20000301 <
US 6509489	B1 20030121	US 2000-694444	20001024 <
WO 2001064618	A1 20010907	WO 2001-US867	20010111 <
W: AL, AM, AT	AU, AZ, BA, BB,	BG, BR, BY, CA, CH, CN	I, CU, CZ, DE,
DK, EE, ES	, FI, GB, GD, GE,	GH, GM, HR, HU, ID, II	, IN, IS, JP,
KE, KG, KP	, KR, KZ, LC, LK,	LR, LS, LT, LU, LV, MI), MG, MK, MN,
MW, MX, NO	NZ, PL, PT, RO,	RU, SD, SE, SG, SI, SH	(, SL, TJ, TM,
TR, TT, UA	, UG, UZ, VN, YU,	ZW, AM, AZ, BY, KG, KZ	Z, MD, RU, TJ, TM
RW: GH, GM, KE	, LS, MW, MZ, SD,	SL, SZ, TZ, UG, ZW, AT	, BE, CH, CY,
DE, DK, ES	, FI, FR, GB, GR,	IE, IT, LU, MC, NL, PT	R, SE, TR, BF,
BJ, CF, CG	CI, CM, GA, GN,	GW, ML, MR, NE, SN, TI), TG
EP 1261578	A1 20021204	EP 2001-901979	20010111 <
R: AT, BE, CH	DE, DK, ES, FR,	GB, GR, IT, LI, LU, NI	, SE, MC, PT,
IE, SI, LT	LV, FI, RO, MK,	CY, AL, TR	
JP 2003525263	T2 20030826	JP 2001-563462	20010111 <
PRIORITY APPLN. INFO.:		US 2000-516746	A3 20000301

. WO 2001-US867 W 20010111

AB The method comprises the step of contacting ≥1 aromatic hydroxy compound with oxygen and CO in the presence of a carbonylation catalyst system having an effective amount of a nickel source as the primary catalyst component and optionally ≥1 inorg. co-catalyst, as well as a halide composition and/or a base in the

absence of a Group VIII B metal source. A process flow diagram is presented.

IT 71-91-0, Tetraethylammonium bromide 1310-73-2,

Sodium hydroxide, uses 14126-37-5

RL: CAT (Catalyst use); USES (Uses)

(carbonylation process and catalyst system for producing diaryl carbonates from the reaction of **carbon monoxide** and **oxygen** with hydroxyarom. compds.)

RN 71-91-0 HCAPLUS

CN Ethanaminium, N,N,N-triethyl-, bromide (9CI) (CA INDEX NAME)

● Br-

RN 1310-73-2 HCAPLUS

CN Sodium hydroxide (Na(OH)) (9CI) (CA INDEX NAME)

Na-OH

RN 14126-37-5 HCAPLUS

CN Nickel, dibromobis(triphenylphosphine) - (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

IT 102-09-0P, Diphenyl carbonate

RL: IMF (Industrial manufacture); PREP (Preparation)
(carbonylation process and catalyst system for producing diaryl
carbonates from the reaction of carbon
monoxide and oxygen with hydroxyarom. compds.)

RN 102-09-0 HCAPLUS

CN Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

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O
||
PhO- C- OPh
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IT 108-95-2, Phenol, reactions 630-08-0,
 Carbon monoxide, reactions 7782-44-7,
 Oxygen, reactions
RL: RCT (Reactant); RACT (Reactant or reagent)
 (carbonylation process and catalyst system for producing diaryl
 carbonates from the reaction of carbon
 monoxide and oxygen with hydroxyarom. compds.)
RN 108-95-2 HCAPLUS

(CA INDEX NAME)

OH

CN

Phenol (8CI, 9CI)

RN 630-08-0 HCAPLUS CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

C -

RN 7782-44-7 HCAPLUS CN Oxygen (8CI, 9CI) (CA INDEX NAME)

o = o

REFERENCE COUNT: 26 THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 13 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

2001:45205 HCAPLUS

DOCUMENT NUMBER:

134:87919

TITLE:

Carbonylation process and catalyst system for producing diaryl carbonates from the reaction of

carbon monoxide and oxygen
with hydroxyaromatic compounds

INVENTOR(S):

Patel, Ben Purushotam; Soloveichik, Grigorii Lev; Whisenhunt, Donald Wayne, Jr.; Shalyaev, Kirill

Vladimirovich

PATENT ASSIGNEE(S):

General Electric Company, USA

SOURCE:

U.S., 6 pp. CODEN: USXXAM

DOCUMENT TYPE:

Patent English

LANGUAGE:

Engi

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

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KIND
                                DATE
                                           APPLICATION NO.
    PATENT NO.
                                                                   DATE
                                           -----
                        _ _ _ _
                                -----
                                20010116 US 2000-510381
20010830 WO 2000-US29285
                         B1
                                                                   20000222 <--
    US 6175033
                         A1
    WO 2001062702
                                                                   20001024 <--
            AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE,
        W:
            DK, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP,
            KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN,
            MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM,
            TR, TT, UA, UG, UZ, VN, YU, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
        RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY,
            DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ,
            CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                         A1 20021204 EP 2000-973807
                                                                   20001024 <--
    EP 1261576
            AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
            IE, SI, LT, LV, FI, RO, MK, CY, AL
                               20020430
                                            US 2000-721682
                         B1
                                                                   20001127 <--
    US 6380418
PRIORITY APPLN. INFO.:
                                            US 2000-510381
                                                                A 20000222
                                            WO 2000-US29285
                                                                W 20001024
    A method of carbonylating aromatic hydroxy compds. into a diaryl carbonate
     (e.g., di-Ph carbonate) comprises reacting at least one aromatic hydroxy
     compound (e.g., phenol) with oxygen and carbon
    monoxide in the presence of a carbonylation catalyst system
     comprising an effective amount of a manganese source [e.g., manganese(II)
     acetylacetonate] as a primary catalyst component in the absence of a Group
     VIIIB metal source, and, optionally in the presence of of a catalytic amount
     of an inorg. cocatalyst [e.g., lead(II) oxide] as well as a halide
     composition (e.g., tetraethylammonium bromide), and/or a base.
     process flow diagram is presented.
IT
     102-09-0P, Diphenyl carbonate
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (carbonylation process and catalyst system for producing diaryl
        carbonates from the reaction of carbon
       monoxide and oxygen with hydroxyarom. compds.)
     102-09-0 HCAPLUS
RN
     Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)
CN
```

IT 108-95-2, Phenol, reactions 630-08-0,
 Carbon monoxide, reactions 7782-44-7,
 Oxygen, reactions
 RL: RCT (Reactant); RACT (Reactant or reagent)
 (carbonylation process and catalyst system for producing diaryl
 carbonates from the reaction of carbon
 monoxide and oxygen with hydroxyarom. compds.)
RN 108-95-2 HCAPLUS
CN Phenol (8CI, 9CI) (CA INDEX NAME)



RN 630-08-0 HCAPLUS

CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

RN 7782-44-7 HCAPLUS

CN Oxygen (8CI, 9CI) (CA INDEX NAME)

o = 0

IT 71-91-0, Tetraethylammonium bromide 1310-73-2,

Sodium hydroxide, uses

RL: CAT (Catalyst use); USES (Uses)

(in a carbonylation catalyst system for producing diaryl carbonates

from the reaction of carbon monoxide and

oxygen with hydroxyarom. compds.)

RN 71-91-0 HCAPLUS

CN Ethanaminium, N,N,N-triethyl-, bromide (9CI) (CA INDEX NAME)

● Br-

RN 1310-73-2 HCAPLUS

CN Sodium hydroxide (Na(OH)) (9CI) (CA INDEX NAME)

· Na-OH

REFERENCE COUNT: 26 THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 14 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

2001:45204 HCAPLUS

DOCUMENT NUMBER:

134:87918

TITLE:

Carbonylation method and catalysts system for

producing diaryl carbonates from the reaction of

carbon monoxide with oxygen and hydroxyaromatic compounds

INVENTOR(S):

Patel, Ben Purushotam; Soloveichik, Grigorii Lev; Whisenhunt, Donald Wayne, Jr.; Shalyaev, Kirill

Vladimirovich

PATENT ASSIGNEE(S):

General Electric Company, USA

SOURCE:

U.S., 6 pp. CODEN: USXXAM

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

m· 1

PATENT INFORMATION:

	PATENT NO.	KIND DATE	APPLICATION NO.	DATE
			US 2000-510380	
			US 2000-665605	
	WO 2001062703		WO 2000-US29312	
			G, BR, BY, CA, CH, CN,	
	•		I, GM, HR, HU, ID, IL,	· · · · · · · · · · · · · · · · · · ·
	•		R, LS, LT, LU, LV, MD,	
			J, SD, SE, SG, SI, SK,	
			, AM, AZ, BY, KG, KZ,	
			S, SZ, TZ, UG, ZW, AT,	
			E, IT, LU, MC, NL, PT,	SE, BF, BJ,
	•		, MR, NE, SN, TD, TG	20002024
	EP 1261577		EP 2000-973814	
			B, GR, IT, LI, LU, NL,	SE, MC, PT,
		LV, FI, RO, MK, CY		20007024
DDTO		T2 20030812	JP 2001-561713 US 2000-510380	
PRIC	RITY APPLN. INFO.:			
3.70	Di	d: Dbb	WO 2000-US29312	
AB			ite) are prepared by c	
		~	e.g., phenol) with oxy	gen and
	carbon monoxide in			[
			nount of a cobalt sour	
			mary catalyst componen	
			llyst [e.g., copper(II	,
	acetylacetonate], a			
T.M.	tetraethylammonium	-	se.	
IT	71-91-0, Tetraethyl			
	RL: CAT (Catalyst u			المسمالة
	-	-	system for producing	diaryi
		the reaction of car		
DM		hydroxyarom. compds	5 . <i>)</i>	
RN	71-91-0 HCAPLUS	t was a short brown i d	(OCT) (CD TAIDEY NAM	TD.)
CN	Ethanaminium, N,N,N	-trietnyi-, bromide	e (9CI) (CA INDEX NAM	E)

• Br-

-OPh Pho-C

TT-108-95-2, Phenol, reactions-630-08-0,

Carbon monoxide, reactions 7782-44-7,

Oxygen, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(carbonylation method and catalysts system for producing diaryl

carbonates from the reaction of carbon

monoxide with oxygen and hydroxyarom. compds.)

RN108-95-2 HCAPLUS

Phenol (8CI, 9CI) CN(CA INDEX NAME)

OH

RN 630-08-0 HCAPLUS

Carbon monoxide (8CI, 9CI) CN (CA INDEX NAME)

RN 7782-44-7 HCAPLUS

CNOxygen (8CI, 9CI) (CA INDEX NAME)

0 = 0

REFERENCE COUNT:

27 THERE ARE 27 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L29 ANSWER 15 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

1997:769987 HCAPLUS

DOCUMENT NUMBER:

128:23258

TITLE:

Process and catalysts for the preparation of diaryl

carbonates from hydroxyaromatic compounds and

carbon monoxide-oxygen gas

mixtures

INVENTOR (S):

Buysch, Hans-Josef; Hesse, Carsten; Rechner, Johann

Bayer A.-G., Germany

PATENT ASSIGNEE(S): SOURCE:

Ger. Offen., 9 pp.

CODEN: GWXXBX

DOCUMENT TYPE:

Patent

LANGUAGE: FAMILY ACC. NUM. COUNT: German

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO.

----------DE 19619949 A1 19971120 DE 1996-19619949 19960517 <--

US 1997-853516 US 5856554 19990105 19970509 <--EP 807619 **A1** 19971119 EP 1997-107407 19970512 <--EP 807619 B120020911 R: BE, DE, ES, FR, GB, IT, NL JP 1997~135803 JP 10045674 A2 19980217 19970512 <--ES 2181946 Т3 20030301 ES 1997-107407 19970512 <--DE 1996-19619949 ·

PRIORITY APPLN. INFO.:

OTHER SOURCE(S):

MARPAT 128:23258

Diaryl carbonates (e.g., di-Ph carbonate) are prepared in high yield, and without the use of phosgene, by the reaction of (un)substituted C6-12 hydroxyarom. compds. (e.g., PhOH) with an O-CO gas mixture in the presence of a platinum-group catalyst (e.g., palladium bromide), a co-catalyst [e.g., manganese(III) acetylacetonate], a quaternary salt (e.g., Bu4NBr), and a base (e.g., PhONa) at 30-200°/1-200 bar in the melt phase and, from the beginning of the reaction, the amount of diaryl carbonate in the reaction mass is maintained at ≥20% (i.e., initially by addition of it to the reaction mixture). A process flow diagram is presented.

1643-19-2, Tetrabutylammonium bromide 13444-94-5,

Palladium bromide

RL: CAT (Catalyst use); USES (Uses)

(process and catalysts for the preparation of diaryl carbonates from hydroxyarom. compds. and carbon monoxide-

oxygen gas mixts.)

1643-19-2 HCAPLUS RN

CN1-Butanaminium, N,N,N-tributyl-, bromide (9CI) (CA INDEX NAME)

● Br -

RN13444-94-5 HCAPLUS CN Palladium bromide (PdBr2) (7CI, 8CI, 9CI) (CA INDEX NAME)

Br-Pd-Br

IT 102-09-0P, Diphenyl carbonate RL: IMF (Industrial manufacture); PREP (Preparation) (process and catalysts for the preparation of diaryl carbonates from hydroxyarom. compds. and carbon monoxide-oxygen gas mixts.)

102-09-0 HCAPLUS RN

CN Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

PhO C OPh

Oxygen, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(process and catalysts for the preparation of diaryl carbonates

from hydroxyarom.compds.and-carbon-monoxide-

oxygen gas mixts.)

RN 108-95-2 HCAPLUS

CN Phenol (8CI, 9CI) (CA INDEX NAME)

OH

RN 630-08-0 HCAPLUS

CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

C +

RN 7782-44-7 HCAPLUS

CN Oxygen (8CI, 9CI) (CA INDEX NAME)

0 = 0

L29 ANSWER 16 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

1997:145244 HCAPLUS

DOCUMENT NUMBER:

126:144050

TITLE:

Preparation of antibiotic and antitumor DC 107

derivatives

INVENTOR(S):

Kanda, Yutaka; Saitoh, Yutaka; Saito, Hiromitsu;

Ashizawa, Tadashi; Sugiyama, Kazuyo; Gomi, Katsushige; Kakita, Shingo; Takahashi, Yuichi; Murakata, Chikara

PATENT ASSIGNEE(S):

Kyowa Hakko Kogyo Co., Ltd., Japan ·

SOURCE:

PCT Int. Appl., 149 pp. CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

NT: 1

PATENT INFORMATION:

PATENT NO.	KIND DATE	APPLICATION NO.	DATE
WO 9700260	A1 19970103	WO 1996-JP1646	19960614 <
W: AU, CA, CN,	HU, JP, KR, NO,	NZ, RU, US	
RW: AT, BE, CH,	DE, DK, ES, FI,	FR, GB, GR, IE, IT, LU,	MC, NL, PT, SE
CA 2197691	AA 19970103	CA 1996-2197691	19960614 <
AU 9660169	A1 19970115	AU 1996-60169	19960614 <
AU 705947	B2 19990603		
EP 786462	A1 19970730	EP 1996-917696	19960614 <
EP 786462	B1 20020918		

R: AT, PT,		DE, DK,	ES, FI,	FR, GB, GR	, IE, IT,	LI, LU,	MC, NL,
CN 1163616		A	19971029	CN 1996	-190920	1	9960614 <
AT 224394		E	20021015	AT 1996	-917696	1	9960614 <
ES 2183958		T3 :	20030401	ES 1996	-917696	1	9960614 <
NO_9700675		Α	19970416	NO 1997	-675	1	9970214 <
NO 309570		B1 :	20010219				
US 5733924		A	19980331	US 1997	-776938	1	9970417 <
PRIORITY APPLN. 1	INFO.:			JP 1995	-150141	A 1	9950616
				WO 1996	-JP1646	W 1	9960614
OTHER SOURCE(S):	1	IARPAT	126:1440	50			

GΙ

DC 107 derivs. I [R1 = H, alkoxyalkyl, aralkyloxyalkyl, alkoxyalkoxyalkyl, AB alkoxyalkoxyalkoxyalkyl, aralkyl, tetrahydropyranyl, COR4, etc.; R2 = H, COR5; R3 = alkyl, alkenyl, (un) substituted aralkyl, etc., R4 = alkyl, etc.; R3 may form a single bond together with Y; Y may form a single bond together with R3 or Z; Z = H or forms a single bond together with Y; W = oxygen, NR6, with provisos] and their pharmaceutically acceptable salts are prepared Thus, DC 107 in CH2Cl2 containing pyridine. acetic anhydride, and 4-dimethylaminopyridine was stirred for 1.5 h to give the title compound II. This had an IC50 of 0.52 mg/mL against Staphylococcus aureus.

IT 186642-77-3P 186643-18-5P 186643-32-3P

> RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(preparation of antibiotic and antitumor DC 107 derivs.)

RN186642-77-3 HCAPLUS

Carbonic acid, (2S,2'R,3R,4R,9'E,11'R,13'E,15'Z)-4-hydroxy-2',4,9'-CN trimethyl-2-oxido-4',5,12'-trioxospiro[1,2-dithiolane-3,6'-

[19]thia[3,20]diazabicyclo[15.2.1]eicosa[1(20),9,13,15,17]pentaen]-11'-yl phenyl ester (9CI) (CA INDEX NAME)

Absolute stereochemistry.

Double bond geometry as described by E or Z.

RN 186643-18-5 HCAPLUS

CN Propanoic acid, 2,2-dimethyl-, [[2-[2,14-dimethyl-20-oxido-11,19-dioxo-12-[(phenoxycarbonyl)oxy]-4,20-dithia-1,21-diazatricyclo[15.2.1.13,6]heneicos a-3(21),5,7,9,13-pentaen-17-yl]-2-hydroxy-1-oxopropyl]thio]methyl ester, [2R,7Z,9E,12R,13E,17R(R),20S]-[partial]- (9CI) (CA INDEX NAME)

Absolute stereochemistry.

Double bond geometry as described by E or Z.

RN 186643-32-3 HCAPLUS

Absolute stereochemistry. Double bond geometry as described by E or Z.

IT 100-11-8, p-Nitrobenzyl bromide 105-36-2, Ethyl bromoacetate 106-95-6, Allyl bromide, reactions 107-30-2 , Chloromethyl methyl ether 109-92-2 116-11-0 501-53-1, Benzyl chloroformate 541-41-3, Ethyl chloroformate 543-27-1, Isobutyl chloroformate 931-57-7 , 1-Methoxy-1-cyclohexene 1885-14-9, Phenyl chloroformate 2687-43-6, O-Benzylhydroxylamine hydrochloride 3188-13-4 , Chloromethyl ethyl ether 3587-60-8, Benzyl chloromethyl ether 3970-21-6 5470-11-1, Hydroxylamine hydrochloride 28920-43-6, 9-Fluorenyl methyl chloroformate 39720-27-9, p-Acetoxybenzyl chloride RL: RCT (Reactant); RACT (Reactant or reagent) (preparation of antibiotic and antitumor DC 107 derivs.) RN 100-11-8 HCAPLUS CN Benzene, 1-(bromomethyl)-4-nitro- (9CI) (CA INDEX NAME)

RN 105-36-2 HCAPLUS CN Acetic acid, bromo-, ethyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

RN 106-95-6 HCAPLUS CN 1-Propene, 3-bromo- (9CI) (CA INDEX NAME)

Br CH2 CH CH2

RN 107-30-2 HCAPLUS

CN Methane, chloromethoxy- (9CI) (CA INDEX NAME)

C1-CH2-O-CH3

RN 109-92-2 HCAPLUS

CN Ethene, ethoxy- (9CI) (CA INDEX NAME)

 $H_3C-CH_2-O-CH-CH_2$

RN 116-11-0 HCAPLUS

CN 1-Propene, 2-methoxy- (9CI) (CA INDEX NAME)

CH₂ || H₃C-O-C-CH₃

RN 501-53-1 HCAPLUS

CN Carbonochloridic acid, phenylmethyl ester (9CI) (CA INDEX NAME)

O || Cl-C-O-CH₂-Ph

RN 541-41-3 HCAPLUS

CN Carbonochloridic acid, ethyl ester (9CI) (CA INDEX NAME)

C1-C-OEt

RN 543-27-1 HCAPLUS

CN Carbonochloridic acid, 2-methylpropyl ester (9CI) (CA INDEX NAME)

O || Cl-C-OBu-i

RN 931-57-7 HCAPLUS

CN Cyclohexene, 1-methoxy- (9CI) (CA INDEX NAME)

OMe

1885-14-9 HCAPLUS RNCarbonochloridic acid, phenyl ester (9CI) (CA INDEX NAME) CN 2687-43-6 HCAPLUS RNHydroxylamine, O-(phenylmethyl)-, hydrochloride (9CI) (CA INDEX NAME) $H_2N-O-CH_2-Ph$ HCl 3188-13-4 HCAPLUS RNEthane, (chloromethoxy) - (9CI) (CA INDEX NAME) H3C-CH2-O-CH2-C1 RN3587-60-8 HCAPLUS Benzene, [(chloromethoxy)methyl] - (9CI) (CA INDEX NAME) ClCH2-O-CH2-Ph 3970-21-6 HCAPLUS RNEthane, 1-(chloromethoxy)-2-methoxy- (7CI, 8CI, 9CI) (CA INDEX NAME) CN $MeO-CH_2-CH_2-O-CH_2Cl$ 5470-11-1 HCAPLUS RN Hydroxylamine, hydrochloride (8CI, 9CI) (CA INDEX NAME)

HCl

 H_2N-OH

RN 28920-43-6 HCAPLUS CN Carbonochloridic acid, 9H-fluoren-9-ylmethyl ester (9CI) (CA INDEX NAME)

39720-27-9 HCAPLUS RN

Phenol, 4-(chloromethyl)-, acetate (9CI) (CA INDEX NAME) CN

L29 ANSWER 17 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

1997:121321 HCAPLUS

DOCUMENT NUMBER:

126:131251

TITLE:

Process and catalysts for the continuous preparation of diaryl carbonates from hydroxyl group-substituted

aromatic compounds

INVENTOR (S):

Buysch, Hans-Josef; Hesse, Carsten; Rechner, Johann

PATENT ASSIGNEE(S):

Bayer A.-G., Germany

SOURCE:

Eur. Pat. Appl., 15 pp.

CODEN: EPXXDW

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.		DATE
EP 749955	A1	19961227	EP 1996-109317		19960611 <
EP 749955	B1	20000816			
R: BE, CH,	DE, ES, FR	, GB, IT,	LI, NL		
DE 19523390	A1	19970109	DE 1995-19523390		19950623 <
US 5625091	A	19970429	US 1996-662431		19960610 <
ES 2151108	Т3	20001216	ES 1996-109317		19960611 <
JP 09012513	A2	19970114	JP 1996-177169	•	19960619 <
CA 2179581	AA	19961224	CA 1996-2179581		19960620 <
CN 1143071	Α	19970219	CN 1996-107161		19960621 <
CN 1119315	В	20030827	<u> </u>		•
PRIORITY APPLN. INFO	.:		DE 1995-19523390	Α	19950623
OTHER SOURCE(S):	MARPAT	126:1312	51		

OTHER SOURCE(S):

The title compds., RO2COR [R = (un) substituted C6-12 aryl] (e.g., di-Ph carbonate), are prepared in a continuous process by the carbonylation of hydroxyl group-substituted aromatic compds. ROH (e.g., PhOH) with CO and 02 in the presence of a Pt-Group metal catalyst (e.g., Pd bromide), a Co catalyst, a quaternary salt (e.g., Bu4NBr), and a base (e.g., PhONa) at 30-200°/1-200 bar, followed by removal of the reaction water under reduced pressure. Process flow diagrams are presented.

1643-19-2, Tetrabutylammonium bromide IT RL: CAT (Catalyst use); USES (Uses) (process and catalysts for the continuous preparation of diaryl carbonates from hydroxyl group-substituted aromatic compds.) 1643-19-2 HCAPLUS RN <u> 1-Butanaminium, N.N.,N-tributyl-, bromide (9CI) (CA_INDEX_NAME)</u> CN_ n-Bu $n-Bu-\dot{N}^{+}Bu-n$ n-Bu · ● Br -102-09-0P, Diphenyl carbonate RL: IMF (Industrial manufacture); SPN (Synthetic preparation); PREP (Preparation) (process and catalysts for the continuous preparation of diary1 carbonates from hydroxyl group-substituted aromatic compds.) RN102-09-0 HCAPLUS Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME) CNPho-C-OPh 108-95-2, Phenol, reactions 630-08-0, IT Carbon monoxide, reactions 7782-44-7, Oxygen, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (process and catalysts for the continuous preparation of diaryl carbonates from hydroxyl group-substituted aromatic compds.) RN108-95-2 HCAPLUS Phenol (8CI, 9CI) (CA INDEX NAME) CNOH 630-08-0 HCAPLUS RN Carbon monoxide (8CI, 9CI) (CA INDEX NAME) CN

RN

7782-44-7 HCAPLUS

Oxygen (8CI, 9CI) (CA INDEX NAME)

0 = 0

```
HCAPLUS COPYRIGHT 2005 ACS on STN
L29 ANSWER 18 OF 24
ACCESSION NUMBER:
                         1997:57544 HCAPLUS
DOCUMENT NUMBER:
                          126:89872
TITLE:
                         Synthesis and characterization of aromatic and
                         brominated aromatic polycarbonates by two-phase
                         phase-transfer-catalyzed polycondensation of
                         bisphenols with trichloromethyl chloroformate
AUTHOR (S):
                         Liaw, Der-Jang; Chang, Ping
CORPORATE SOURCE:
                         Dep. Chem. Eng., Natl. Taiwan Inst. Technol., Taipei,
                         106, Taiwan
SOURCE:
                         Journal of Applied Polymer Science (1997),
                          63(2), 195-204
                         CODEN: JAPNAB; ISSN: 0021-8995
PUBLISHER:
                         Wiley
DOCUMENT TYPE:
                         Journal
LANGUAGE:
                         English
     Aromatic and brominated aromatic homo polycarbonates were synthesized by the
     two-phase phase-transfer-catalyzed polycondensation of bisphenols with
     trichloromethyl chloroformate at 25°C. The IR spectra, inherent
     viscosity, x-ray diffraction, solubility, contact angle, differential scanning
     calorimetry, thermogravimetric anal., and limiting oxygen index
     (LOI) of all polycarbonates were measured. Polycarbonates of moderate or
     large molar mass with inherent viscosities up to 0.77 dL/g were obtained
     in high yields with tetrabutylammonium bromide (TBAB) as a catalyst,
     sodium hydroxide as a base, and
     1,2-dichloroethane as solvent. The brominated polycarbonates have good
     flame retardancy, as indicated by LOI values. The x-ray diffraction
     diagram showed that all polycarbonates were semicryst. The polycarbonate
     (PC-2) based on bisphenol S has greater crystallinity than the others
     because of the sulfonyl group, which has a small van der Waals radius.
     The incorporation of the bromine atoms (PC-4-PC-6) on the ring decreased
     the crystallinity. Almost all polymers were soluble in DMF, pyridine, and phenol, but insol. in acetone and m-cresol. Solubility increased remarkably
     with bromine substitution. The contact angles of polycarbonates
     (PC-1-PC-3) lie in the of range 82 to 97 degrees greater than that of
     brominated polycarbonates (PC-4-PC-6). The wettability of the homo
     polycarbonate based on bisphenol S is greater than that of
     polycarbonates-derived from bisphenol A and bisphenol AF.
     polycarbonates lies in the range 141-206AC, although Tg of
     polycarbonate based on bisphenol S was not detected Tg of brominated
     polycarbonates was remarkably greater than that of unbrominated
     polycarbonates. These polymers obtained from aromatic bisphenols lost no
     mass below 341°C, but 10% loss of mass was recorded above
     396°C in nitrogen.
IT
     1643-19-2, Tetrabutylammonium bromide 5197-95-5,
     Benzyltriethylammonium bromide
     RL: CAT (Catalyst use); USES (Uses)
        (synthesis and characterization of aromatic and brominated aromatic
        polycarbonates by two-phase phase-transfer-catalyzed polycondensation
        of bisphenols with trichloromethyl chloroformate)
```

1-Butanaminium, N,N,N-tributyl-, bromide (9CI) (CA INDEX NAME)

1643-19-2 HCAPLUS

RN

CN

Br-

5197-95-5 HCAPLUS RN

Benzenemethanaminium, N,N,N-triethyl-, bromide (9CI) (CA INDEX NAME) CN

 Et_3+N-CH_2-Ph

● Br⁻

24936-68-3P, preparation 28774-93-8P, IT

3,3',5,5'-Tetrabromobisphenol A-trichloromethyl chloroformate copolymer, sru 28930-33-8P, Bisphenol S-trichloromethyl chloroformate copolymer, sru 32291-26-2P, Bisphenol AF-trichloromethyl chloroformate copolymer, sru 56912-08-4P, 3,3',5,5'-

Tetrabromobisphenol S-trichloromethyl chloroformate copolymer, sru 126430-95-3P, 3,3',5,5'-Tetrabromobisphenol AF-trichloromethyl chloroformate copolymer, sru

RL: PRP (Properties); SPN (Synthetic preparation); PREP

(Preparation)

(synthesis and characterization of aromatic and brominated aromatic polycarbonates by two-phase phase-transfer-catalyzed polycondensation of bisphenols with trichloromethyl chloroformate)

RN24936-68-3 HCAPLUS

Poly[oxycarbonyloxy-1,4-phenylene(1-methylethylidene)-1,4-phenylene] (9CI) CN (CA INDEX NAME)

RN28774-93-8 HCAPLUS

Poly[oxycarbonyloxy(2,6-dibromo-1,4-phenylene)(1-methylethylidene)(3,5-CN dibromo-1,4-phenylene)] (9CI) (CA INDEX NAME)

RN 28930-33-8 HCAPLUS

CN Poly(oxycarbonyloxy-1,4-phenylenesulfonyl-1,4-phenylene) (9CI) (CA INDEX NAME)

RN 32291-26-2 HCAPLUS

CN Poly[oxycarbonyloxy-1,4-phenylene[2,2,2-trifluoro-1-(trifluoromethyl)ethylidene]-1,4-phenylene] (9CI) (CA INDEX NAME)

RN 56912-08-4 HCAPLUS

CN Poly[oxycarbonyloxy(2,6-dibromo-1,4-phenylene)sulfonyl(3,5-dibromo-1,4-phenylene)] (9CI) (CA INDEX NAME)

RN 126430-95-3 HCAPLUS

CN Poly[oxycarbonyloxy(2,6-dibromo-1,4-phenylene)[2,2,2-trifluoro-1-(trifluoromethyl)ethylidene](3,5-dibromo-1,4-phenylene)] (9CI) (CA INDEX NAME)

IT 80-09-1 1478-61-1

RL: RCT (Reactant); RACT (Reactant or reagent)
(synthesis and characterization of aromatic and brominated aromatic polycarbonates by two-phase phase-transfer-catalyzed polycondensation of bisphenols with trichloromethyl chloroformate)

RN 80-09-1 HCAPLUS

CN Phenol, 4,4'-sulfonylbis- (9CI) (CA INDEX NAME)

RN 1478-61-1 HCAPLUS

CN Phenol, 4,4'-[2,2,2-trifluoro-1-(trifluoromethyl)ethylidene]bis- (9CI) (CA INDEX NAME)

IT 39635-79-5P, 3,3',5,5'-Tetrabromobisphenol S

RL: RCT (Reactant); SPN (Synthetic preparation); PREP

(Preparation); RACT (Reactant or reagent)

(synthesis and characterization of aromatic and brominated aromatic polycarbonates by two-phase phase-transfer-catalyzed polycondensation of bisphenols with trichloromethyl chloroformate)

RN 39635-79-5 HCAPLUS

CN Phenol, 4,4'-sulfonylbis[2,6-dibromo- (9CI) (CA INDEX NAME)

REFERENCE COUNT:

THERE ARE 45 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

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L29 ANSWER 19 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN
ACCESSION NUMBER:
                        1996:674344 HCAPLUS
DOCUMENT NUMBER:
                        125:300614
TITLE:
                        Process and catalysts for the preparation of diaryl
                        -carbonates-from-aryl-alcohols-and-carbon-----
                        monoxide and oxygen
                        Buysch, Hans-Josef; Hesse, Carsten; Rechner, Johann
INVENTOR(S):
PATENT ASSIGNEE(S):
                        Bayer A.-G., Germany
SOURCE:
                        Eur. Pat. Appl., 10 pp.
                        CODEN: EPXXDW
DOCUMENT TYPE:
                        Patent
                        German
LANGUAGE:
FAMILY ACC. NUM. COUNT:
PATENT INFORMATION:
                             DATE
                                         APPLICATION NO.
                        KIND
     PATENT NO.
                                                                DATE
                        ----
                                          -----
                               -----
                                                                 ------
                                        EP 1996-104710
    EP 736511
                        A1
                               19961009
                                                                 19960325 <--
    EP 736511
                        B1
                               19991027
        R: DE, ES, FR, GB, IT, NL
    DE 19512616 A1
                               19961010
                                          DE 1995-19512616
                                                                 19950405 <--
     ES 2139269
                        Т3
                               20000201
                                           ES 1996-104710
                                                                 19960325 <--
                                           JP 1996-95861
     JP 08283206
                        A2
                               19961029
                                                                 19960327 <--
                                           US 1996-623728
     US 5663408
                               19970902
                                                                 19960329 <--
                        Α
PRIORITY APPLN. INFO.:
                                           DE 1995-19512616
                                                             A 19950405
OTHER SOURCE(S):
                        MARPAT 125:300614
    Diaryl carbonates ROCO2R [R = (un) substituted C6-12 aryl] (e.g., di-Ph
    carbonate) are prepared in high yield and selectivity by the reaction of
     aryl alcs. ROH (e.g., PhOH) with CO and O2 at
     30-200°/2-50 bars in the presence of a quaternary salt (e.g.,
     Bu4NBr), a base (e.g., PhONa), a Pt-group metal catalyst (e.g.,
     palladium bromide), and a co-catalyst [e.g., activated C and Mn
     (II) acetylacetonate].
IT
     102-09-0P, Diphenyl carbonate
     RL: IMF (Industrial manufacture); SPN (Synthetic preparation); PREP
     (Preparation)
        (process and catalysts for the preparation of diaryl
        carbonates from aryl alcs. and carbon
       monoxide and oxygen)
     102-09-0 HCAPLUS
RN
     Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)
CN
Pho-C-OPh
     108-95-2, Phenol, reactions 630-08-0,
IT
     Carbon monoxide, reactions 1643-19-2,
     Tetrabutylammonium bromide 7782-44-7, Oxygen,
     reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (process and catalysts for the preparation of diaryl carbonates
        from aryl alcs. and carbon monoxide and
        oxygen)
     108-95-2 HCAPLUS
RN
     Phenol (8CI, 9CI) (CA INDEX NAME)
CN
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RN 630-08-0 HCAPLUS

CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

C | ↓

RN 1643-19-2 HCAPLUS

CN 1-Butanaminium, N,N,N-tributyl-, bromide (9CI) (CA INDEX NAME)

● Br-

RN 7782-44-7 HCAPLUS

CN Oxygen (8CI, 9CI) (CA INDEX NAME)

o = 0

L29 ANSWER 20 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

1995:774826 HCAPLUS

DOCUMENT NUMBER:

123:169244

TITLE:

Process for continuous preparation of diaryl

carbonates

INVENTOR(S):

Buysch, Hans-Josef; Hesse, Carsten; Rechner, Johann; Schomaecker, Reinhard; Wagner, Paul; Kaufmann, Dieter

Prof Dipl Chem

PATENT ASSIGNEE(S):

Bayer A.-G., Germany

SOURCE:

Ger. Offen., 10 pp.

CODEN: GWXXBX

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 4403075	A1	19950803	DE 1994-4403075	19940202 <
EP 667336	A1	19950816	EP 1995-100787	19950120 <
EP 667336	. B1	19980520		

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R: BE, CH, DE, ES, FR, GB, IT, LI, NL
                                19980816 ES 1995-100787
     ES 2117808
                          T3
                                                                    19950120 <--
     JP 07247243
                          A2
                                19950926
                                            JP 1995-31483
                                                                    19950127 <--
     US 5498742
                          Α
                                19960312
                                            US 1995-379384
                                                                    19950127 <--
     CA 2141391
                          AA
                                19950803
                                            CA 1995-2141391
                                                                    19950130 <--
     CN-1112107-
                          A.
                                19951122
                                            -CN-1995-101656-
                                                                    -19950130-<--
     CN 1056365
                          В
                                20000913
PRIORITY APPLN. INFO.:
                                            DE 1994-4403075
                                                                 A 19940202
OTHER SOURCE(S):
                         CASREACT 123:169244; MARPAT 123:169244
     Improvements are made in the preparation of diaryl carbonates (RO)2CO [R =
     (un) substituted C6-12 aryl] by reaction of phenols ROH with CO
     and O2 in the presence of a CO-activated noble metal
     catalyst (group VIIIb), a cocatalyst, a quaternary salt, and a
     base. In particular, the reaction is conducted with removal of
     H2O by stripping of the reaction mixture with excess reaction gas.
     example, a run was performed at 80° with 450 g PhOH, with PdBr2 as
     catalyst, Mn(II) acetylacetonate as cocatalyst, NaOPh as base,
     and in the presence of Bu4N+ Br-. The reaction gas was a (95:5) mixture of
     CO and O2 at 10 bar, introduced at a rate of 400 NL/h.
     The reaction mixture had a content of 18.6% (PhO) 2CO after 3 h, with removal
     of 8.75 g PhOH-H2O mixture as condensate. In contrast, a non-invention run
     using only 6 NL/h gas mixture gave only 5.4% (PhO)2CO content in 3 h, with
     only 0.2 g condensate.
IT
     13444-94-5, Palladium dibromide
     RL: CAT (Catalyst use); USES (Uses)
        (catalyst; continuous preparation of diaryl carbonates)
RN
     13444-94-5 HCAPLUS
     Palladium bromide (PdBr2) (7CI, 8CI, 9CI) (CA INDEX NAME)
CN
Br-Pd-Br
     102-09-0P, Diphenyl carbonate
     RL: IMF (Industrial manufacture); SPN (Synthetic preparation); PREP
     (Preparation)
        (continuous preparation of diaryl carbonates)
RN
     102-09-0 HCAPLUS
CN
     Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)
Pho-c-oph
     108-95-2, Phenol, reactions 630-08-0,
     Carbon monoxide, reactions 7782-44-7,
     Oxygen, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reactant; continuous preparation of diaryl carbonates)
RN
     108-95-2 HCAPLUS
     Phenol (8CI, 9CI) (CA INDEX NAME)
CN
```

RN 630-08-0 HCAPLUS

CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

) | | |

RN 7782-44-7 HCAPLUS

CN Oxygen (8CI, 9CI) (CA INDEX NAME)

o = o

IT 1643-19-2, Tetrabutylammonium bromide

RL: NUU (Other use, unclassified); USES (Uses)

(reaction component; continuous preparation of diaryl carbonates)

RN 1643-19-2 HCAPLUS

CN 1-Butanaminium, N,N,N-tributyl-, bromide (9CI) (CA INDEX NAME)

• Br-

L29 ANSWER 21 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

1993:652566 HCAPLUS

DOCUMENT NUMBER:

119:252566

TITLE:

Manufacture of aromatic carbonates

INVENTOR(S):

Joyce, Richard P.; King, Joseph A., Jr.; Pressman,

Eric J.

PATENT ASSIGNEE(S):

General Electric Co., USA

SOURCE:

U.S., 5 pp.

DOCUMENT TYPE:

CODEN: USXXAM Patent

LANGUAGE

English

LANGUAGE:

Engii

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PA	TENT NO.		KIND	DATE	APPLICATION NO.	DATE
US	5231210		Α	19930727	US 1992-929749	19920817 <
EF	583935		A1	19940223	EP 1993-306328	19930811 <
EF	583935		B1	19961120		
	R: DE	, ES, FF	R, GB, IT	C, NL		
ES	2094485		Т3	19970116	ES 1993-306328	19930811 <
JE	06172268	3	A2	19940621	JP 1993-202161	19930816 <
JE	2971297		B2	19991102		
PRIORIT	Y APPLN.	INFO.:			US 1992-929749 [.]	A 19920817

AB The process comprises heating to a temperature of 60-150° a mixture comprising an aromatic hydroxy compound, CO, O, and an effective amount of a Pd carbonylation catalyst comprising (a) catalytically active Pd in the metallic or chemical bonded state, (b) an inorg. cocatalyst in the form of a complex of a Co2+ salt and a Schiff base, (c) a quaternary ammonium-or-phosphonium-halide,-and-(d)-optionally,-aterpyridine compound This catalyst combination substantially enhances the production rate as compared to similar catalysts containing Co2+ and Mn3+ salts instead of the Co2+-Schiff base complex. An stirred autoclave containing a mixture of PhOH 36.41 and (Bu) 4NBr 1.16118 g, and Pd(OAc) 2 26.8, terpyridine 9.6, and Co di(salicylal)-3,3'-diamino-Nmethyldipropylamine (I) 24.6 mg, and (Ph)20 5.01 g was flushed with CO at 400 psi, pressurized with O to 110 psi and with CO to 590 psi, and heated at 115° to give Ph2CO3 at 0.35 mol/L.h, vs. 0.17 with Co(OAc)2 instead of I. IT1643-19-2, Tetrabutylammonium bromide RL: CAT (Catalyst use); USES (Uses) (catalysts, for carbonylation of phenols with carbon monoxide and oxygen) RN 1643-19-2 HCAPLUS CN 1-Butanaminium, N,N,N-tributyl-, bromide (9CI) (CA INDEX NAME) n-Bu n-Bu-h+ Bu-n n-Bu ● Br-102-09-0P, Diphenyl carbonate RL: IMF (Industrial manufacture); PREP (Preparation) (manufacture of, catalysts for) RN 102-09-0 HCAPLUS CNCarbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME) Pho-c-oph IT 108-95-2, Phenol, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (reaction of, with carbon monoxide and oxygen in manufacture of di-Ph carbonate, catalysts for) RN 108-95-2 HCAPLUS Phenol (8CI, 9CI) (CA INDEX NAME) CN

7782-44-7, Oxygen, reactions IT RL: RCT (Reactant); RACT (Reactant or reagent) (reaction of, with phenol and carbon monoxide in manufacture of aromatic carbonates, catalysts for) 7782-44-7 HCAPLUS RN Oxygen (8CI, 9CI) (CA INDEX NAME) o = 0630-08-0, Carbon monoxide, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (reaction of, with phenols and oxygen in manufacture of aromatic carbonates, catalysts for) 630-08-0 HCAPLUS RNCarbon monoxide (8CI, 9CI) (CA INDEX NAME) CN L29 ANSWER 22 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN ACCESSION NUMBER: 1993:449073 HCAPLUS 119:49073 DOCUMENT NUMBER: Preparation of aromatic carbonates TITLE: ' Fujita, Terunori; Kiso, Yoshihisa; Nagata, Takuji; INVENTOR(S): Iwasaki, Hiroshi Mitsui Petrochemical Industries, Co., Ltd., Japan PATENT ASSIGNEE(S): Jpn. Kokai Tokkyo Koho, 4 pp. SOURCE: CODEN: JKXXAF Patent DOCUMENT TYPE: Japanese LANGUAGE: FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION: APPLICATION NO. DATE KIND DATE PATENT NO. ---------· ________ 19930202 JP 1991~203636 19910719 <--A2 JP 05025095 B2 20000228 JP 3014812 JP 1991-203636 19910719 PRIORITY APPLN. INFO.: Aromatic carbonates are prepared by treatment of aromatic hydroxy compds. with CO and mol. O in the presence of catalysts comprising (a) Pd and/or Pd compds., (b) Co compds., (c) organic and/or inorg. halides, and (d) bases under pressure and heating. Autoclaving phenol, Pd(OAc)2, Co(OPh)2, CsBr, and Cs2CO3 at 100° and CO 49, 0 25, and CO2 15 kg/cm2 for 3 h gave 4.6% di-Ph carbonate in 96% selectivity, vs. 0.3% and 72%, without CsBr, resp. 106-44-5, p-Cresol, reactions 108-95-2, ITPhenol, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (carbonylation of, with carbon monoxide and oxygen)

106-44-5 HCAPLUS

Phenol, 4-methyl- (9CI) (CA INDEX NAME)

RN

CN

RN 108-95-2 HCAPLUS

CN Phenol (8CI, 9CI) (CA INDEX NAME)

IT 1310-73-2, Sodium hydroxide, uses

1643-19-2, Tetrabutylammonium bromide 7758-02-3,

Potassium bromide, uses 7787-69-1, Cesium bromide

RL: USES (Uses)

(catalyst systems containing, in carbonylation of phenols)

RN 1310-73-2 HCAPLUS

CN Sodium hydroxide (Na(OH)) (9CI) (CA INDEX NAME)

Na-OH

RN 1643-19-2 HCAPLUS

CN 1-Butanaminium, N,N,N-tributyl-, bromide (9CI) (CA INDEX NAME)

● Br-

RN 7758-02-3 HCAPLUS

CN Potassium bromide (KBr) (9CI) (CA INDEX NAME)

Br-K

RN 7787-69-1 HCAPLUS

CN Cesium bromide (CsBr) (9CI) (CA INDEX NAME)

Br—Cs

IT 534-17-8, Cesium carbonate 584-08-7, Potassium

carbonate 584-09-8, Rubidium carbonate

RL: RCT (Reactant); RACT (Reactant or reagent)

(catalyst systems containing, in carbonylation of phenols)

RN 534-17-8 HCAPLUS

CN Carbonic acid, dicesium salt (8CI, 9CI) (CA INDEX NAME)

о || но-с-он

•2 Cs

RN 584-08-7 HCAPLUS

CN Carbonic acid, dipotassium salt (8CI, 9CI) (CA INDEX NAME)

о || но— с— он

●2 K

RN 584-09-8 HCAPLUS

CN Carbonic acid, dirubidium salt (8CI, 9CI) (CA INDEX NAME)

но-с-он

•2 Rb

IT 102-09-0P, Diphenyl carbonate

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of, from phenol and carbon monoxide and oxygen)

RN 102-09-0 HCAPLUS

CN Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

O || PhO- C- OPh

IT 621-02-3P, Bis(p-tolyl) carbonate 33524-49-1P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of, from phenol derivative and carbon monoxide and oxygen)

RN 621-02-3 HCAPLUS

Carbonic acid, bis(4-methylphenyl) ester (9CI) (CA INDEX NAME) CN

RN 33524-49-1 HCAPLUS

Phenol, 4-(1-methyl-1-phenylethyl)-, carbonate (2:1) (9CI) (CA INDEX CN

HCAPLUS COPYRIGHT 2005 ACS on STN L29 ANSWER 23 OF 24

ACCESSION NUMBER:

1980:604290 HCAPLUS

DOCUMENT NUMBER:

93:204290

TITLE:

Aromatic carbonates Hallgren, John Edward

INVENTOR(S): PATENT ASSIGNEE(S):

General Electric Co., USA

SOURCE:

Ger. Offen., 8 pp. CODEN: GWXXBX

DOCUMENT TYPE:

Patent

LANGUAGE:

German

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

	PATENT NO.	KIND	DATE .	APPLICATION NO.	DATE
	DD 0040006			DE 4050 004004	
	DE 2949936	A1	19800703	DE 1979-2949936	19791212 <
	GB 2038321	Α	19800723	GB 1979-39487	19791114 <
	GB 2038321	B2	19830413		
	JP 55102539	A2	19800805	JP 1979-160409	19791212 <
	NL 7908991	A	19800617	NL 1979-8991	19791213 <
	FR 2444024	A1	19800711	FR 1979-30668	19791214 <
	CA 1137102	A1	19821207	CA 1979-342001	19791214 <
PRIOF	RITY APPLN. INFO.:			US 1978-969546	19781214
AB	Aromatic carbonates	are pre	epared by tre	eating a phenol with a	n alkanol and
				Mn redox cocatalyst,	
				CHC6H4C6H4OH-4 was tre	
	with EtOH and CO in	_	·		
				ctivated Linde 3A mol.	sieve to give
				1C6H4O)2CO, and (EtO)2	
IT		JC 1, 1	(4 NOSCHOOM	1001110,200, and (ECO)2	
11	13444-94-5				
	RL: RCT (Reactant);	RACT (F	Reactant or 1	reagent)	

(catalyst for reaction of carbon monoxide with phenols and alkanols)

13444-94-5 HCAPLUS RN

CNPalladium bromide (PdBr2) (7CI, 8CI, 9CI) (CA INDEX NAME) Br-Pd-Br

IT 102-09-0P 75422-89-8P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

RN 102-09-0 HCAPLUS

CN Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

RN 75422-89-8 HCAPLUS

CN [1,1'-Biphenyl]-4-ol, 4'-(1-methylethyl)-, carbonate (2:1) (9CI) (CA INDEX NAME)

IT 1643-19-2

RN

RL: RCT (Reactant); RACT (Reactant or reagent)
 (reaction of carbon monoxide with phenols
 and alkanols in presence of)

1643-19-2 HCAPLUS

CN 1-Butanaminium, N,N,N-tributyl-, bromide (9CI) (CA INDEX NAME)

• Br-

IT 108-95-2, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)
 (reaction of, with methanol and carbon monoxide)

RN 108-95-2 HCAPLUS

CN Phenol (8CI, 9CI) (CA INDEX NAME)

Sackey 10 687411 IT 630-08-0, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (reaction of, with phenols and alkanols, aromatic carbonates from) RN630-08-0 HCAPLUS -Carbon-monoxide-(8CI,-9CI)--(CA-INDEX-NAME)-----CN-L29 ANSWER 24 OF 24 HCAPLUS COPYRIGHT 2005 ACS on STN ACCESSION NUMBER: 1980:146429 HCAPLUS DOCUMENT NUMBER: 92:146429 Catalytic preparation of aromatic carbonates TITLE: PATENT ASSIGNEE(S): General Electric Co., USA Fr. Demande, 19 pp. Addn. to Fr. Demande 2,367,731. SOURCE: CODEN: FRXXBL DOCUMENT TYPE: Patent French LANGUAGE: FAMILY ACC. NUM. COUNT: PATENT INFORMATION: PATENT NO. KIND DATE APPLICATION NO. DATE --------------FR 2422621 A2 19791109 FR 1978-10542 19780410 <--FR 2422621 B2 19820122 PRIORITY APPLN. INFO.: FR 1978-10542 A 19780410 Aromatic carbonates were prepared by the reaction of phenols with CO, an oxidant and a base in the presence of a catalyst chosen from Ru, Rh, Pd, Os, Ir or Pt or their compds. and optionally with one of a variety of metal compound cocatalysts. Thus, p-PhCMe2C6H4OH treated 44 h with CO in the presence of 1,2,2,6,6-pentamethylpiperidine, PdBr2 and Mn(ON:CPhCHPhOH)2 gave 96% conversion of phenol with formation of 95 mol carbonate per mol PdBr2. With bisphenol A the polycarbonate was formed. IT 7787-70-4 13446-03-2 13470-26-3 RL: CAT (Catalyst use); USES (Uses) (catalyst, for oxidative reaction of phenols with carbon monoxide) RN 7787-70-4 HCAPLUS CN Copper bromide (CuBr) (8CI, 9CI) (CA INDEX NAME)

Br--- Cu

RN 13446-03-2 HCAPLUS CN Manganese bromide (MnBr2) (6CI, 8CI, 9CI) (CA INDEX NAME)

Br-Mn-Br

RN 13470-26-3 HCAPLUS CN Vanadium bromide (VBr3) (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

IT 80-05-7, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (oxidative reaction of, with carbon monoxide, polycarbonates by)

RN 80-05-7 HCAPLUS

CN Phenol, 4,4'-(1-methylethylidene)bis- (9CI) (CA INDEX NAME)

630-08-0, reactions IT RL: RCT (Reactant); RACT (Reactant or reagent) (oxidative reaction of, with phenols, aromatic carbonates by)

RN 630-08-0 HCAPLUS

· CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

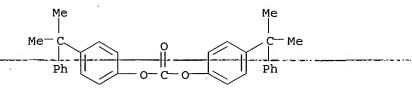
IT 24936-68-3P, preparation 33524-49-1P RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

RN24936-68-3 HCAPLUS

Poly[oxycarbonyloxy-1,4-phenylene(1-methylethylidene)-1,4-phenylene] (9CI) CN (CA INDEX NAME)

RN 33524-49-1 HCAPLUS

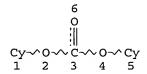
CN Phenol, 4-(1-methyl-1-phenylethyl)-, carbonate (2:1) (9CI) NAME)



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L1 STR



NODE ATTRIBUTES:

DEFAULT MLEVEL IS ATOM DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RING(S) ARE ISOLATED OR EMBEDDED

NUMBER OF NODES IS

STEREO ATTRIBUTES: NONE

DIDKEO.	ATIKIDUI	BO. NONE
L2		SEA FILE=REGISTRY SSS FUL L1
L3	125595	SEA FILE=REGISTRY ABB=ON PLU=ON ACTIVATING (W) SOLVENT OR
		ETHER? OR SULFONE? OR NITRILES OR AMIDES OR CARBONATE? OR
		POLYETHER? OR DIGLYME OR TRIGLYME OR TETRAGLYME
L4 ·	1255	SEA FILE=REGISTRY ABB=ON PLU=ON SOLVENT OR SOLVENTS
L5	95	SEA FILE=REGISTRY ABB=ON PLU=ON NITRILE?/CN
L6	786	SEA FILE=REGISTRY ABB=ON PLU=ON AMIDE?/CN
L7	16418	SEA FILE=REGISTRY ABB=ON PLU=ON PHENOLIC OR CRESOL OR
		4-FLUOROPHENOL?/CN OR BISPHENOL A?/CN OR METHYL SALICYLATE?/CN
L8		SEA FILE=REGISTRY ABB=ON PLU=ON PHENOL/CN
L9	24018	SEA FILE=HCAPLUS ABB=ON PLU=ON L2 OR DIARYL(W)CARBONATE
L10	2183760	SEA FILE=HCAPLUS ABB=ON PLU=ON L3 OR L4 OR L5 OR L6 OR
		ACTIVATING (W) SOLVENT OR ETHER? OR SULFONE? OR NITRILE OR AMIDE
		OR CARBONATE? OR POLYETHER? OR DIGLYME OR TRIGLYME OR
		TETRAGLYME
L11	570160	SEA FILE=HCAPLUS ABB=ON PLU=ON L7 OR L8 OR PHENOLIC OR
		CRESOL OR 4 (W) FLUOROPHENOL? OR BISPHENOL (W) A OR METHYL (W)
		SALICYLATE? OR PHENOL
L12	5461	SEA FILE=HCAPLUS ABB=ON PLU=ON L9(L)PREPARATION/RL
L13	335274	SEA FILE=HCAPLUS ABB=ON PLU=ON REACTANT/RL(L)L10
L14	65329	SEA FILE=HCAPLUS ABB=ON PLU=ON REACTANT/RL(L)L11
L15	677	SEA FILE=HCAPLUS ABB=ON PLU=ON L12 AND L13 AND L14
L16	101866	SEA FILE=REGISTRY ABB=ON PLU=ON PALLADIUM OR ACETYLACETONATE
L17	19213	SEA FILE=REGISTRY ABB=ON PLU=ON CARBON MONOXIDE?/CN OR
		OXYGEN
L19	17811	SEA FILE=REGISTRY ABB=ON PLU=ON (TETRAMETHYLAMMONIUM OR
		TETRAMETHYL (L) AMMONIUM OR PHOSPHONIUM OR AMMONIUM OR LITHIUM
		OR SODIUM OR POTASSIUM) (L) HYDROXIDE OR (AMINE OR TRIETHYLAMIN
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E OR TRIALKYLAMINE) (L) HYDRATE
L20
         165455 SEA FILE=REGISTRY ABB=ON PLU=ON HALIDE OR BROMIDE OR
                (LITHIUM OR MAGNESIUM) (L) BROMIDE OR (AMMONIUM OR PHOSPHONIUM) (W
                ) HALIDE OR ALKALI METAL?/CN
         197938 SEA FILE=HCAPLUS ABB=ON PLU=ON L16 OR PALLADIUM OR ACETYLACET
L22
                ONATE
        1846505 SEA FILE=HCAPLUS ABB=ON PLU=ON L17 OR CARBON (W) MONOXIDE OR
L23
                CO OR OXYGEN OR O2
         904208 SEA FILE=HCAPLUS ABB=ON PLU=ON L19 OR BASE OR (PHOSPHONIUM
L24
                OR ?AMMONIUM OR LITHIUM OR SODIUM OR POTASSIUM) (3A) HYDROXIDE
                OR ?AMINE (5A) HYDRATE
L25
         565021 SEA FILE=HCAPLUS ABB=ON PLU=ON L20 OR HALIDE OR BROMIDE
                ALKALI (W) METAL?
L26
         243271 SEA FILE=REGISTRY ABB=ON PLU=ON
                                                  COPPER?/CN
L27
         170458 SEA FILE=REGISTRY ABB=ON PLU=ON
                                                  TITANIUM
L28
             25 SEA FILE=HCAPLUS ABB=ON PLU=ON L15 AND L23 AND L24 AND L25
                                                 L28 AND PD=<OCTOBER 14, 2003
L29
             24 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON
             38 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON
                                                 L13 AND L14 AND L23 AND L22
L30
                AND L24 AND L25
             18 SEA FILE=HCAPLUS ABB=ON
                                                 L30 NOT L29
L31
                                         PLU=ON
L32
              5 SEA FILE=HCAPLUS ABB=ON
                                         PLU=ON
                                                 L31 AND (L26 OR L27 OR
                CO(W) CATALY?)
L33
              5 SEA FILE=HCAPLUS ABB=ON PLU=ON
                                                  (L32 OR L28) NOT L29
=>
=>
=> d ibib abs hitstr 133 1-5
L33 ANSWER 1 OF 5 HCAPLUS COPYRIGHT 2005 ACS on STN
ACCESSION NUMBER:
                         2005:406839 HCAPLUS
                           Correction of: 2005:155216
                           Correction of: 142:197768
                         Product class 1: pyridines
TITLE:
AUTHOR(S):
                         Spitzner, D.
CORPORATE SOURCE:
                         Germany
                         Science of Synthesis (2005), 15, 11-284
SOURCE:
                         CODEN: SSCYJ9
PUBLISHER:
                         Georg Thieme Verlag
DOCUMENT TYPE:
                         Journal; General Review
LANGUAGE:
                         English
     A review of methods to prepare pyridines, pyridine-1-oxides, and pyridinium
     salts. Methods include cyclization, ring transformations, aromatization
     and substituent modification.
IT
     INDEXING IN PROGRESS
     142-71-2 506-68-3, Cyanogen bromide ((CN)Br)
     506-96-7, Acetyl bromide 544-92-3, Copper cyanide
     (Cu(CN)) 576-83-0 1310-65-2, Lithium
     hydroxide (Li(OH)) 1336-21-6, Ammonium
     hydroxide ((NH4)(OH)) 1643-19-2 2857-97-8
     3375-31-3 5470-11-1 7550-35-8, Lithium bromide
     (LiBr) 7550-45-0, Titanium chloride (TiCl4) (T-4)-
     7647-10-1, Palladium chloride (PdCl2) 7647-15-6
      Sodium bromide (NaBr) 7699-45-8, Zinc bromide (ZnBr2)
     7720-78-7 7727-15-3, Aluminum bromide (AlBr3)
     7758-89-6, Copper chloride (CuCl) 7789-47-1, Mercury
     bromide (HqBr2) 7789-59-5, Phosphoric tribromide
     10028-15-6, Ozone 10035-10-6, Hydrobromic acid
```

13965-03-2 14221-01-3 26323-01-3

29964-62-3 51364-51-3

RL: CAT (Catalyst use); USES (Uses)

(review of preparation of pyridines, pyridine-1-oxides and pyridinium salts via cyclization, ring transformations, aromatization and substituent modification)

-RN---142--71-2---HCAPLUS-----

CN Acetic acid, copper(2+) salt (8CI, 9CI) (CA INDEX NAME)

●1/2 Cu(II)

RN 506-68-3 HCAPLUS

CN Cyanogen bromide ((CN)Br) (9CI) (CA INDEX NAME)

Br-C = N

RN 506-96-7 HCAPLUS

CN Acetyl bromide (6CI, 8CI, 9CI) (CA INDEX NAME)

RN 544-92-3 HCAPLUS

CN Copper cyanide (Cu(CN)) (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

RN 576-83-0 HCAPLUS

CN Benzene, 2-bromo-1,3,5-trimethyl- (9CI) (CA INDEX NAME)

RN 1310-65-2 HCAPLUS

CN Lithium hydroxide (Li(OH)) (9CI) (CA INDEX NAME)

Li-OH

RN 1336-21-6 HCAPLUS

CN Ammonium hydroxide ((NH4)(OH)) (9CI) (CA INDEX NAME)

 H_4N-OH

RN 1643-19-2 HCAPLUS

CN 1-Butanaminium, N,N,N-tributyl-, bromide (9CI) (CA INDEX NAME)

• Br-

RN · 2857-97-8 HCAPLUS

CN Silane, bromotrimethyl- (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

RN 3375-31-3 HCAPLUS

CN Acetic acid, palladium(2+) salt (8CI, 9CI) (CA INDEX NAME)

●1/2 Pd(II)

RN 5470-11-1 HCAPLUS

CN Hydroxylamine, hydrochloride (8CI, 9CI) (CA INDEX NAME)

 H_2N-OH

● HCl

RN 7550-35-8 HCAPLUS

CN Lithium bromide (LiBr) (9CI) (CA INDEX NAME)

Br-Li

RN 7550-45-0 HCAPLUS

----CN Titanium chloride (TiCl4) -- (T-4) -- (9CI) (CA INDEX NAME)

RN 7647-10-1 HCAPLUS

CN Palladium chloride (PdCl2) (6CI, 8CI, 9CI) (CA INDEX NAME)

Cl-Pd-Cl

RN 7647-15-6 HCAPLUS

CN Sodium bromide (NaBr) (9CI) (CA INDEX NAME)

Br-Na

RN 7699-45-8 HCAPLUS

CN Zinc bromide (ZnBr2) (9CI) (CA INDEX NAME)

Br-Zn-Br

RN 7720-78-7 HCAPLUS

CN Sulfuric acid, iron(2+) salt (1:1) (8CI, 9CI) (CA INDEX NAME)

• Fe(II)

RN 7727-15-3 HCAPLUS

CN Aluminum bromide (AlBr3) (9CI) (CA INDEX NAME)

RN 7758-89-6 HCAPLUS

CN Copper chloride (CuCl) (8CI, 9CI) (CA INDEX NAME)

Cl-Cu

RN 7789-47-1 HCAPLUS

CN Mercury bromide (HgBr2) (8CI, 9CI) (CA INDEX NAME)

Br-Hg-Br

RN 7789-59-5 HCAPLUS

CN Phosphoric tribromide (9CI) (CA INDEX NAME)

RN 10028-15-6 HCAPLUS

CN Ozone (8CI, 9CI) (CA INDEX NAME)

0-0-0

RN 10035-10-6 HCAPLUS

CN Hydrobromic acid (8CI, 9CI) (CA INDEX NAME)

HBr

RN 13965-03-2 HCAPLUS

CN Palladium, dichlorobis(triphenylphosphine) - (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

RN 14221-01-3 HCAPLUS

CN Palladium, tetrakis(triphenylphosphine)-, (T-4)- (9CI) (CA INDEX NAME)

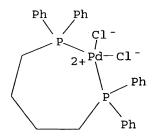
RN26323-01-3 HCAPLUS

CN Quinolinium, 1-(phenylmethyl)-, bromide (9CI) (CA INDEX NAME)

● Br‐

RN

29964-62-3 HCAPLUS
Palladium, [1,4-butanediylbis[diphenylphosphine-κP]]dichloro-, (SP-4-2)- (9CI) (CA INDEX NAME) CN



RN51364-51-3 HCAPLUS

Palladium, tris $[\mu - [(1,2-\eta:4,5-\eta)-(1E,4E)-1,5-diphenyl-1,4-$ CNpentadien-3-one]]di- (9CI) (CA INDEX NAME)

IT 67-64-1, 2-Propanone 75-16-1 100-58-3 106-96-7 108-24-7 108-95-2, Phenol 112-71-0 122-51-0 124-63-0, Methanesulfonyl chloride 135-02-4 623-00-7 2259-30-5 13058-25-8 13735-81-4 17015-31-5 21970-14-9 24424-99-5 34896-80-5 61049-69-2 62479-73-6 73296-31-8 171926-14-0 189001-08-9 367906-47-6 RL: RCT (Reactant); RACT (Reactant or reagent) (review of preparation of pyridines, pyridine-1-oxides and pyridinium salts via cyclization, ring transformations, aromatization and substituent modification) RN 67-64-1 HCAPLUS CN 2-Propanone (9CI) (CA INDEX NAME)

RN 75-16-1 HCAPLUS CN Magnesium, bromomethyl- (8CI, 9CI) (CA INDEX NAME)

Br-Mg-CH3

RN 100-58-3 HCAPLUS CN Magnesium, bromophenyl- (8CI, 9CI) (CA INDEX NAME)

Ph-Mq-Br

RN 106-96-7 HCAPLUS

CN 1-Propyne, 3-bromo- (9CI) (CA INDEX NAME)

 $Br-CH_2-C \equiv CH$

RN 108-24-7 HCAPLUS

CN Acetic acid, anhydride (9CI) (CA INDEX NAME)

Ac- 0- Ac

RN 108-95-2 HCAPLUS

CN Phenol (8CI, 9CI) (CA INDEX NAME)

RN 112-71-0 HCAPLUS

CN Tetradecane, 1-bromo- (6CI, 7CI, 8CI, 9CI) (CA INDEX NAME)

 $Me^{-(CH_2)_{13}-Br}$

RN 122-51-0 HCAPLUS

CN Ethane, 1,1',1''-[methylidynetris(oxy)]tris- (9CI) (CA INDEX NAME)

RN 124-63-0 HCAPLUS

CN Methanesulfonyl chloride (6CI, 8CI, 9CI) (CA INDEX NAME)

RN 135-02-4 HCAPLUS

CN Benzaldehyde, 2-methoxy- (9CI) (CA INDEX NAME)

RN 623-00-7 HCAPLUS

CN Benzonitrile, 4-bromo- (9CI) (CA INDEX NAME)

RN 2259-30-5 HCAPLUS

CN Magnesium, bromo(1,1-dimethylethyl) - (9CI) (CA INDEX NAME)

t-Bu-Mg-Br

RN 13058-25-8 HCAPLUS

CN Cyclopropylium, 1,2,3-triphenyl-, bromide (8CI, 9CI) (CA INDEX NAME)

• Br

RN 13735-81-4 HCAPLUS

CN Silane, trimethyl[(1-phenylethenyl)oxy] - (9CI) (CA INDEX NAME)

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Ph-C-O-SiMe}_3 \end{array}$$

RN 17015-31-5 HCAPLUS

CN 1,3-Butadiene, 1-ethoxy-2-methyl-, (1E)- (9CI) (CA INDEX NAME)

Double bond geometry as shown.

RN 21970-14-9 HCAPLUS

CN Magnesium, bromo-3-pyridinyl- (9CI) (CA INDEX NAME)

RN 24424-99-5 HCAPLUS

CN Dicarbonic acid, bis(1,1-dimethylethyl) ester (9CI) (CA INDEX NAME)

RN 34896-80-5 HCAPLUS

CN Benzene, 1-bromo-3-(methylsulfonyl)- (9CI) (CA INDEX NAME)

RN 61049-69-2 HCAPLUS

CN 4H-Pyran-4-one, 2-methyl-3-(phenylmethoxy)- (9CI) (CA INDEX NAME)

RN 62479-73-6 HCAPLUS

CN Formaldehyde, O-methyloxime (7CI, 9CI) (CA INDEX NAME)

 $H_2C = N - O - CH_3$

RN 73296-31-8 HCAPLUS

CN Silane, trimethyl(phenyltelluro) - (9CI) (CA INDEX NAME)

Me₃Si-Te-Ph

RN 171926-14-0 HCAPLUS

CN Pyridinium, 4-(dimethoxymethyl)-1-[2-(1H-indol-3-yl)ethyl]-, bromide (9CI) (CA INDEX NAME)

Sackey 10_687411

$$\begin{array}{c|c} & \text{OMe} \\ & \text{CH} - \text{OMe} \\ & \\ & \text{CH}_2 - \text{CH}_2 - \text{N} \end{array}$$

🗎 Bri

RN 189001-08-9 HCAPLUS
CN Pyridinium, 1-[2-[4-[2-(1,3-dioxolan-2-yl)ethenyl][2,2':4',4'':2'',2''':4'
'',4''''-quinquepyridin]-2''''-yl]-2-oxoethyl]-, bromide (9CI) (CA INDEX NAME)

PAGE 1-A

O CH CH CH CH C CH2

● Br~

PAGE 1-B



RN 367906-47-6 HCAPLUS
CN Pyridinium, 1-(phenylmethyl)-3-[(2-thioxo-3-thiazolidinyl)carbonyl]-,
 bromide (9CI) (CA INDEX NAME)

● Br -

IT 109-04-6P 626-55-1P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(review of preparation of pyridines, pyridine-1-oxides and pyridinium salts via cyclization, ring transformations, aromatization and substituent modification)

RN 109-04-6 HCAPLUS

CN Pyridine, 2-bromo- (6CI, 8CI, 9CI) (CA INDEX NAME)

RN 626-55-1 HCAPLUS

CN Pyridine, 3-bromo- (6CI, 8CI, 9CI) (CA INDEX NAME)

IT 74569-95-2P

RL: SPN (Synthetic preparation); PREP (Preparation) (review of preparation of pyridines, pyridine-1-oxides and pyridinium salts via cyclization, ring transformations, aromatization and substituent modification)

RN 74569-95-2 HCAPLUS

CN Spiro[cyclopentane-1,4'(1'H)-[2,7]naphthyridinium], 2',3'-dihydro-1',3'-dioxo-7'-(phenylmethyl)-, bromide (9CI) (CA INDEX NAME)

Br-

L33 ANSWER 2 OF 5 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER: 2005:349044 HCAPLUS

DOCUMENT NUMBER: 142:394138

TITLE:

Water-resistant carbonylation catalyst system for the

production of diaryl carbonates via the direct

carbonylation of phenolic compounds

Soloveichik, Grigorii Lev; Chuck, Timothy Leigh; INVENTOR(S):

Shalyaev, Kirill Vladimirovich; Pressman, Eric James;

Bonitatebus, Peter John

PATENT ASSIGNEE(S):

SOURCE:

General Electric Company, USA U.S. Pat. Appl. Publ., 9 pp.

CODEN: USXXCO

DOCUMENT TYPE:

LANGUAGE:

Patent English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.				KIND		DATE		APPLICATION NO.						DATE				
					-													
US 2005085656				A1		20050421		US 2003-687411						20031015				
WO :	WO 2005040089				A2		20050506		WO 2004-US30610						20040917			
	W:	ΑE,	AG,	AL,	AM,	AT,	AU,	ΑZ,	BA,	BB,	BG,	BR,	BW,	BY,	ΒZ,	CA,	CH,	
		CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,	GB,	GD,	
		GE,	GH,	GM,	HR,	ΗU,	ID,	IL,	IN,	IS,	JP,	KE,	KG,	KΡ,	KR,	ΚZ,	LC,	
		LK,	LR,	LS,	LT,	LU,	LV,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,	NA,	NI,	
		NO,	NZ,	OM,	PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,	SK,	SL,	SY,	
		TJ,	TM,	TN,	TR,	TT,	TZ,	UA,	ŪĠ,	US,	UZ,	VC,	VN,	YU,	ZA,	ZM,	ZW	
	RW:	BW,	GH,	GM,	KE,	LS,	MW,	MZ,	NA,	SD,	SL,	SZ,	TZ,	UG,	ZM,	ZW,	AM,	
		AZ,	BY,	KG,	ΚZ,	MD,	RU,	TJ,	TM,	ΑT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,	
		EE,	ES,	FI,	FR,	GB,	GR,	HU,	ΙE,	IT,	LU,	MC,	NL,	ΡL,	PT,	RO,	SE,	
		SI,	SK,	TR,	BF,	ВJ,	CF,	CG,	CI,	CM,	GΑ,	GN,	GQ,	GW,	ML,	MR,	NE,	
		SN,	TD,	TG														

PRIORITY APPLN. INFO.:

US 2003-687411 A 20031015

OTHER SOURCE(S): CASREACT 142:394138

A method of increasing the amount of diaryl carbonates (e.g., di-Ph carbonate) produced per amount of catalyst consumed in a phenolic compound (e.g., phenol) carbonylation process is described. Phenolic compound carbonylation produces water as a reaction byproduct which reduces the turnover number (TON) of the catalyst. A mixture of a phenolic precursor, a base-containing catalyst and co-catalyst components and at least one chemical additive comprising a halide or hydroxide of alkali metal or alkaline earth metal when carbonylated together under specific conditions increases the TON and water resistivity of a palladium catalyst. The metal halide likely

Sackey 10 687411

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makes the catalyst less susceptible to degradation by water hence increasing
     the reaction yield per weight of catalyst consumed.
IT
     75-59-2, Tetramethylammonium hydroxide
     1310-58-3, Potassium hydroxide, uses
     1310-65-2, Lithium hydroxide 1310-73-2
    --, Sodium hydroxide, uses 7440-05-3,
     Palladium, uses 7440-32-6, Titanium, uses
     7440-50-8, Copper, uses 7550-35-8, Lithium bromide
     7789-48-2, Magnesium bromide 14024-61-4,
     Palladium acetylacetonate 27143-60-8,
     Triethylamine monohydrate
     RL: CAT (Catalyst use); USES (Uses)
        (in a water-resistant carbonylation catalyst system for the production of
        diaryl carbonates via the direct carbonylation of phenolic compds.)
     75-59-2 HCAPLUS
RN
     Methanaminium, N,N,N-trimethyl-, hydroxide (9CI) (CA INDEX NAME)
CN
    CH<sub>3</sub>
    CH3
  ● OH -
     1310-58-3 HCAPLUS
RN
     Potassium hydroxide (K(OH)) (9CI) (CA INDEX NAME)
CN
K-OH
     1310-65-2 HCAPLUS
RN
     Lithium hydroxide (Li(OH)) (9CI) (CA INDEX NAME)
CN
Li-OH
     1310-73-2 HCAPLUS
RN
     Sodium hydroxide (Na(OH)) (9CI) (CA INDEX NAME)
CN
Na-OH
     7440-05-3 HCAPLUS
RN
     Palladium (8CI, 9CI) (CA INDEX NAME)
CN
Pd
RN
     7440-32-6 HCAPLUS
     Titanium (8CI, 9CI)
                          (CA INDEX NAME)
CN
```

Тi

RN 7440-50-8 HCAPLUS
CN Copper (7CI, 8CI, 9CI) (CA INDEX NAME)

Cu

RN 7550-35-8 HCAPLUS

CN Lithium bromide (LiBr) (9CI) (CA INDEX NAME)

Br-Li

RN 7789-48-2 HCAPLUS

CN Magnesium bromide (MgBr2) (9CI) (CA INDEX NAME)

Br-Mg-Br

RN 14024-61-4 HCAPLUS

CN Palladium, bis(2,4-pentanedionato- κ 0, κ 0')-, (SP-4-1)- (9CI) (CA INDEX NAME)

RN 27143-60-8 HCAPLUS

CN Ethanamine, N, N-diethyl-, monohydrate (9CI) (CA INDEX NAME)

● H₂O

IT 630-08-0, Carbon monoxide, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(in a water-resistant carbonylation catalyst system for the production of diaryl carbonates via the direct carbonylation of

phenolic compds.)

RN 630-08-0 HCAPLUS

CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

C-0+

IT 7782-44-7, Oxygen, reactions

RL: RGT (Reagent); RACT (Reactant or reagent)

(in a water-resistant carbonylation catalyst system for the production of diaryl carbonates via the direct carbonylation of phenolic compds.)

RN 7782-44-7 HCAPLUS

CN Oxygen (8CI, 9CI) (CA INDEX NAME)

o = o

IT 102-09-0P, Diphenyl carbonate

RL: IMF (Industrial manufacture); PREP (Preparation)
(water-resistant carbonylation catalyst system for the production of diaryl carbonates via the direct carbonylation of phenolic compds.)

RN 102-09-0 HCAPLUS

CN Carbonic acid, diphenyl ester (6CI, 8CI, 9CI) (CA INDEX NAME)

0 || PhO-- C-- OPh

IT 80-05-7, Bisphenol A, reactions 95-48-7, o-Cresol, reactions 106-44-5, p-

Cresol, reactions 108-39-4, m-Cresol, reactions 108-95-2, Phenol, reactions 119-36-8

, Methyl salicylate 371-41-5, 4-

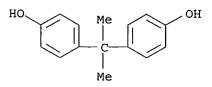
Fluorophenol

RL: RCT (Reactant); RACT (Reactant or reagent)

(water-resistant carbonylation catalyst system for the production of diaryl carbonates via the direct carbonylation of **phenolic** compds.)

RN 80-05-7 HCAPLUS

CN Phenol, 4,4'-(1-methylethylidene)bis- (9CI) (CA INDEX NAME)



RN 95-48-7 HCAPLUS

CN Phenol, 2-methyl- (9CI) (CA INDEX NAME)

RN 106-44-5 HCAPLUS

CN Phenol, 4-methyl- (9CI) (CA INDEX NAME)

RN 108-39-4 HCAPLUS

CN Phenol, 3-methyl- (9CI) (CA INDEX NAME)

RN .108-95-2 HCAPLUS

CN Phenol (8CI, 9CI) (CA INDEX NAME)

RN 119-36-8 HCAPLUS

CN Benzoic acid, 2-hydroxy-, methyl ester (9CI) (CA INDEX NAME)

RN 371-41-5 HCAPLUS

CN Phenol, 4-fluoro- (9CI) (CA INDEX NAME)

Sackey 10_687411

L33 ANSWER 3 OF 5 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER: 1991:228664 HCAPLUS

DOCUMENT NUMBER: 114:228664

-TITLE: Synthesis of cycloprop[c]indol-5-ones from

4-diazo-3-[n-(2-propenyl)amido]cyclohexadien-1-ones. Exploration of copper(I) and copper(II) complexes as

catalysts

AUTHOR(S): Sundberg, Richard J.; Pitts, William J.

CORPORATE SOURCE: Dep. Chem., Univ. Virginia, Charlottesville, VA,

22901, USA

SOURCE: Journal of Organic Chemistry (1991), 56(9), 3048-54

CODEN: JOCEAH; ISSN: 0022-3263

Ι

DOCUMENT TYPE: Journal LANGUAGE: English

OTHER SOURCE(S): CASREACT 114:228664

GI

$$\begin{array}{c} N_2 \\ N \text{ (CH}_2\text{CH}:\text{CH}_2\text{) SO}_2\text{Me} \\ \\ \text{Me} \\ N_2 \\ \text{NRCH}_2\text{CH}:\text{CH}_2 \end{array}$$

AB The cyclization of diazo(propenylamino)cyclohexadienones (I) and (II; R = SO2Me, COMe) to cyclopropindolones (III) and (IV) under the influence of Cu(I) and Cu(II) compds. has been investigated. Catalysis is observed with Cu(I) triflate, the CO complex of Cu(I) triflate, and the CO complexes of trifluoropentanedionato- and hexafluoropentanedionatoCu(I). The best results, essentially quant. conversion, are achieved with a catalyst solution containing trifluoropentanedionato Cu(I) carbonyl and 1 equiv of BuNH2. No significant enantioselectivity is observed with a chiral salicyliminato Cu(II), [(trifluoroacetyl)camphorato Cu(I) carbonyl, or a trifluoropentanedionato Cu(I) carbonyl solution containing (S)- α -

IV

Sackey 10_687411

naphthylethylamine. A mechanistic interpretation involving reductive dediazonization, exo-trig radical cyclization, and cyclopropane formation by the resulting intermediate is proposed.

IT 7440-05-3, Palladium, uses and miscellaneous

RL: CAT (Catalyst use); USES (Uses)

(catalysts, for hydrogenation of nitro(propenylacetamido)indole)

RN 7440-05-3 HCAPLUS

CN Palladium (8CI, 9CI) (CA INDEX NAME)

Pd

IT 1643-19-2, Tetrabutylammonium bromide

RL: RCT (Reactant); RACT (Reactant or reagent)
(cyclization of bromonitropropenylaniline in presence of, triethylamine and palladium acetate)

RN 1643-19-2 HCAPLUS

CN 1-Butanaminium, N,N,N-tributyl-, bromide (9CI) (CA INDEX NAME)

• Br-

IT 3375-31-3

RL: RCT (Reactant); RACT (Reactant or reagent) (cyclization of bromonitropropenylaniline in presence of, triethylamine and tetrabutylammonium bromide)

RN 3375-31-3 HCAPLUS

CN Acetic acid, palladium(2+) salt (8CI, 9CI) (CA INDEX NAME)

●1/2 Pd(II)

IT 7789-45-9, Copper dibromide

RL: PROC (Process)

(cyclization of diazocyclohexadienone in presence of)

RN 7789-45-9 HCAPLUS

CN Copper bromide (CuBr2) (6CI, 8CI, 9CI) (CA INDEX NAME)

Br-Cu-Br

IT 10294-33-4, Boron tribromide

Sackey 10 687411

RL: RCT (Reactant); RACT (Reactant or reagent) (isomerization of nitroindole derivative with)

RN 10294-33-4 HCAPLUS

CN Borane, tribromo- (9CI) (CA INDEX NAME)

RN. 7705-07-9 HCAPLUS

CN Titanium chloride (TiCl3) (8CI, 9CI) (CA INDEX NAME)

IT 81967-72-8P 95345-21-4P 95345-22-5P 133471-93-9P 133471-94-0P 133574-98-8P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of, as catalyst for cyclopropanation of diazo(propenylamino)cyclohexadienone derivs.)

RN 81967-72-8 HCAPLUS

CN Copper, carbonyl(trifluoromethanesulfonato- κ O) - (9CI) (CA INDEX NAME)

RN 95345-21-4 HCAPLUS

CN Copper, carbonyl(1,1,1,5,5,5-hexafluoro-2,4-pentanedionato-0,0')- (9CI) (CA INDEX NAME)

RN 95345-22-5 HCAPLUS

CN Copper, carbonyl(1,1,1-trifluoro-2,4-pentanedionato-0,0')- (9CI) (CA INDEX NAME)

RN 133471-93-9 HCAPLUS

CN Copper, (1-butanamine) carbonyl (1,1,1-trifluoro-2,4-pentanedionato-0,0')-, (T-4)- (9CI) (CA INDEX NAME)

Me
$$CF_3$$

NH₂-Bu-n

Cu+

CF₃

RN 133471-94-0 HCAPLUS

CN Copper, carbonyl(α -methyl-1-naphthalenemethanamine)(1,1,1-trifluoro-2,4-pentanedionato-0,0')-, [T-4-(S)]- (9CI) (CA INDEX NAME)

RN 133574-98-8 HCAPLUS

CN Copper, carbonyl[1,7,7-trimethyl-3-(trifluoroacetyl)bicyclo[2.2.1]heptan-2-onato-0,0']- (9CI) (CA INDEX NAME)

RN 106-95-6 HCAPLUS

CN 1-Propene, 3-bromo- (9CI) (CA INDEX NAME)

 $Br-CH_2-CH-CH_2$

IT 42152-44-3

RL: RCT (Reactant); RACT (Reactant or reagent)
 (reaction of, with carbon monoxide and
 hexafluoropentanedione)

RN 42152-44-3 HCAPLUS

CN Methanesulfonic acid, trifluoro-, copper(1+) salt (9CI) (CA INDEX NAME)

Cu(I)

IT 630-08-0, Carbon monoxide, reactions
RL: RCT (Reactant); RACT (Reactant or reagent)

(reaction of, with cuprous oxide and trifluoromethanesulfonic acid)

RN 630-08-0 HCAPLUS

CN Carbon monoxide (8CI, 9CI) (CA INDEX NAME)

C-

IT 106-45-6

RL: RCT (Reactant); RACT (Reactant or reagent)
 (reaction of, with diazocyclohexadienone)

RN 106-45-6 HCAPLUS

CN Benzenethiol, 4-methyl- (9CI) (CA INDEX NAME)

541-41-3, Ethyl chloroformate IT

RL: RCT (Reactant); RACT (Reactant or reagent)

(reaction of, with nitro(propenylacetamido)indole)

RN 541-41-3 HCAPLUS

Carbonochloridic acid, ethyl ester (9CI) (CA INDEX NAME) CN

Cl-C-OEt

IT

1317-39-1, Cuprous oxide, reactions
RL: RCT (Reactant); RACT (Reactant or reagent)

(reaction of, with trifluoromethanesulfonic acid and carbon

monoxide) 1317-39-1 HCAPLUS RN

Copper oxide (Cu2O) (8CI, 9CI) (CA INDEX NAME) CN

Cu-o-cu

L33 ANSWER 4 OF 5 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

1987:515455 HCAPLUS

DOCUMENT NUMBER:

107:115455

TITLE:

Copper-catalyzed double cyclization reaction of

azidoquinones: one-step synthesis of dihydropyrroloindoloquinones and related

quinolinoquinones

AUTHOR (S):

Naruta, Yoshinori; Nagai, Naoshi; Arita, Yoshihiro;

Maruyama, Kazuhiro

CORPORATE SOURCE:

Fac. Sci., Kyoto Univ., Kyoto, 606, Japan

SOURCE:

Journal of Organic Chemistry (1987), 52(18), 3956-67

II

CODEN: JOCEAH; ISSN: 0022-3263

DOCUMENT TYPE:

Journal

LANGUAGE:

English

OTHER SOURCE(S):

CASREACT 107:115455

GI

MeO CHRCH= CHCH=
$$CH_2$$
 Me N_3

Sackey 10 687411

AB Intramol. cyclization of azido(pentadienyl)quinone I (R = H) has been examined in the presence of metal salts, e.g., MLn (M = Cu, Mn, Co , etc.; L = acetylacetonato). Copper or CuL2 exhibited the highest catalytic activity both to the decomposition of the azide and to the formation of the corresponding dihydropyrroloindoloquinone II (R = H) which was obtained in 58% yield in one step. The related azido(hexadienyl)quinones gave the corresponding quinolinoquinone derivs. in moderate yields. pyrolysis of hexadienylquinone derivative I (R = Me) in benzene in the presence of CuL2 afforded 27% quinolinoquinone. The double cyclization reaction proceeds with extremely high regio- and stereoselectivity, and the generality was established. Quinonoid structures and the presence of a conjugated dienyl side chain at the proximal position to an azide group are essential for the completion of this double cyclization reaction. The role of the copper catalyst to the cyclization reaction is also discussed. IT 1560-54-9, Allyltriphenylphosphonium bromide RL: RCT (Reactant); RACT (Reactant or reagent)

(allylation of (formylethyl)benzene derivative with lithiated derivative of) 1560-54-9 HCAPLUS

RN

Phosphonium, triphenyl-2-propenyl-, bromide (9CI) (CA INDEX NAME) CN

 $Ph_3+P-CH_2-CH=CH_2$

Br-

IT 13395-16-9

> RL: CAT (Catalyst use); USES (Uses) (catalysts, for thermal decomposition and ring closure of azido alkadienyl

RN13395-16-9 HCAPLUS

Copper, bis $(2,4-pentanedionato-\kappa 0,\kappa 0')$ -, (SP-4-1)- (9CI)CN INDEX NAME)

(CA INDEX NAME)

ΙT 3153-26-2, Oxobis (acetylacetonato) vanadium 3264-82-2, Bis (acetylacetonato) nickel 13476-99-8, Tris(acetylacetonato) vanadium 14024-18-1, Tris(acetylacetonato)iron 14024-48-7, Bis(acetylacetonato)cobalt 14024-58-9, Bis (acetylacetonato) manganese 21679-31-2, Tris(acetylacetonato)chromium RL: CAT (Catalyst use); USES (Uses) (catalysts, for thermal decomposition of azidopentadienyl quinone) RN3153-26-2 HCAPLUS Vanadium, oxobis(2,4-pentanedionato-κ0,κ0')-, (SP-5-21)- (9CI) CN

$$\begin{array}{c|cccc}
Me & Me \\
\hline
HC & V2+ & CH \\
\hline
Me & O & Me
\end{array}$$

RN 3264-82-2 HCAPLUS

CN Nickel, bis(2,4-pentanedionato- κ 0, κ 0')-, (SP-4-1)- (9CI) (CA INDEX NAME)

RN . 13476-99-8 HCAPLUS

CN Vanadium, tris(2,4-pentanedionato-κ0,κ0')-, (OC-6-11)- (9CI)
 (CA INDEX NAME)

RN 14024-18-1 HCAPLUS

CN Iron, tris(2,4-pentanedionato-κ0,κ0')-, (OC-6-11)- (9CI) (CA
INDEX NAME)

RN 14024-48-7 HCAPLUS
CN Cobalt, bis(2,4-pentanedionato-κ0,κ0')-, (SP-4-1)- (9CI) (CAINDEX NAME)

RN 14024-58-9 HCAPLUS
CN Manganese, bis(2,4-pentanedionato-κO,κO')- (9CI) (CA INDEX NAME)

Copper bromide (CuBr) (8CI, 9CI) (CA INDEX NAME)

CN

Br-Cu

RN 14024-63-6 HCAPLUS
CN Zinc, bis(2,4-pentanedionato-κ0,κ0')-, (T-4)- (9CI) (CA INDEX NAME)

RN 14040-05-2 HCAPLUS
CN Copper, bis(2,2,6,6-tetramethyl-3,5-heptanedionato-κ0,κ0')(9CI) (CA INDEX NAME)

RN 14128-84-8 HCAPLUS
CN Copper, bis(1-phenyl-1,3-butanedionato-κ0,κ0')- (9CI) (CA INDEX NAME)

RN 14167-15-8 HCAPLUS

CN Copper, [[2,2'-[1,2-ethanediylbis[(nitrilo-κN)methylidyne]]bis[phenolato-κO]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

RN 14172-91-9 HCAPLUS

CN Copper, [5,10,15,20-tetraphenyl-21H,23H-porphinato(2-)kN21,kN22,kN23,kN24]-, (SP-4-1)- (9CI) (CA INDEX NAME)

RN 14221-10-4 HCAPLUS

CN Copper, bis[[2,3-butanedione di(oximato- κ N)](1-)]-, (SP-4-1)- (9CI) (CA INDEX NAME)

RN 14263-53-7 HCAPLUS

CN Copper, [[4,4'-[1,2-ethanediyldi(nitrilo- κ N)]bis[2-pentanonato- κ O]](2-)]-, (SP-4-2)- (9CI) (CA INDEX NAME)

RN 14284-06-1 HCAPLUS

CN Copper, bis[ethyl 3-(οχο-κΟ)butanoato-κΟ']- (9CI) (CA INDEX NAME)

EtO OEt

OCU
$$^{2+}$$
 CH

Me Me

RN 14284-89-0 HCAPLUS

CN Manganese, tris(2,4-pentanedionato- κ 0, κ 0')-, (OC-6-11)- (9CI) (CA INDEX NAME)

RN 14324-82-4 HCAPLUS

CN Copper, bis(1,1,1-trifluoro-2,4-pentanedionato-κ0,κ0')- (9CI) (CA INDEX NAME)

RN 14405-48-2 HCAPLUS

RN 14523-25-2 HCAPLUS

CN Copper, bis [2-(hydroxy- κ O) benzaldehydato- κ O] - (9CI) (CA INDEX NAME)

RN 14781-49-8 HCAPLUS

CN Copper, bis(3-methyl-2,4-pentanedionato- κ 0, κ 0')- (9CI) (CA INDEX NAME)

RN 21679-46-9 HCAPLUS

CN Cobalt, tris(2,4-pentanedionato- κ O, κ O')-, (OC-6-11)- (9CI) (CA INDEX NAME)

RN 42152-44-3 HCAPLUS

CN Methanesulfonic acid, trifluoro-, copper(1+) salt (9CI) (CA INDEX NAME)

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- SO3H
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Cu(I)

107-30-2, Chloromethyl methyl ether IT

> RL: RCT (Reactant); RACT (Reactant or reagent) (methoxymethylation by, of substituted phenols)

RN107-30-2 HCAPLUS

Methane, chloromethoxy- (9CI) (CA INDEX NAME) CN

C1-CH2-O-CH3

L33 ANSWER 5 OF 5 HCAPLUS COPYRIGHT 2005 ACS on STN

ACCESSION NUMBER:

1987:33289 HCAPLUS

DOCUMENT NUMBER:

106:33289

TITLE:

E and C parameters from Hammett substituent constants and use of E and C to understand cobalt-carbon bond

energies

AUTHOR (S):

SOURCE:

Drago, Russell S.; Wong, Ngai; Bilgrien, Carl; Vogel,

CORPORATE SOURCE:

Chem. Dep., Univ. Florida, Gainesville, FL, 32611, USA

Inorganic Chemistry (1987), 26(1), 9-14

CODEN: INOCAJ; ISSN: 0020-1669

DOCUMENT TYPE:

Journal LANGUAGE: English

An updated list of E and C parameters was calculated from a larger data base than that used in an earlier fit. The new data base included 42 acids, 55 bases, and about 500 data points. Best-fit parameters for 13 enthalpy-frequency shift relations were also reported. From this updated list, relationships were discovered which lead to equations that enabled calcn. of E and C parameters from Hammett substituent consts. for a series of substituted phenols and pyridines. This procedure provided a simple method for greatly increasing the number of acids and bases included in the correlation. An E and C anal. was used to study the dissociation energy of the Co-C bond in alkyl-substituted bis(dimethylglyoximato)cobalt(II) complexes. gave calculated dissociation energies that were within exptl. error of the measured values and gave a value for Co-C bond dissociation for the unligated complex. The basic procedure allows for the incorporation of ligand influence on bond dissociation energies into the correlation.

115-10-6, Dimethyl ether 123-91-1, Dioxane, TT

properties 141-78-6, Ethyl acetate, properties 142-96-1

, Dibutyl ether 150-19-6, m-Methoxyphenol

150-76-5, p-Methoxyphenol 352-93-2, Diethyl sulfide

371-41-5, p-Fluorophenol 626-55-1, 3-Bromopyridine

629-82-3, Di-n-octyl ether 7789-33-5

12081-18-4 14781-45-4, Bis(hexafluoroacetylacetonato)cop

per(II)

RL: RCT (Reactant); RACT (Reactant or reagent)

Sackey 10_687411

(electrostatic and covalent parameters of)

RN 115-10-6 HCAPLUS

CN Methane, oxybis- (9CI) (CA INDEX NAME)

Н3С-0-СН3

RN 123-91-1 HCAPLUS

CN 1,4-Dioxane (9CI) (CA INDEX NAME)

RN 141-78-6 HCAPLUS

CN Acetic acid ethyl ester (8CI, 9CI) (CA INDEX NAME)

Et-O-Ac

RN 142-96-1 HCAPLUS

CN Butane, 1,1'-oxybis- (9CI) (CA INDEX NAME)

n-Bu-O-Bu-n

RN 150-19-6 HCAPLUS

CN Phenol, 3-methoxy- (9CI) (CA INDEX NAME)

HOOMe

RN 150-76-5 HCAPLUS

CN Phenol, 4-methoxy- (9CI) (CA INDEX NAME)

HOOMe

RN 352-93-2 HCAPLUS

CN Ethane, 1,1'-thiobis- (9CI) (CA INDEX NAME)

RN 371-41-5 HCAPLUS

CN Phenol, 4-fluoro- (9CI) (CA INDEX NAME)

RN 626-55-1 HCAPLUS

CN Pyridine, 3-bromo- (6CI, 8CI, 9CI) (CA INDEX NAME)

RN 629-82-3 HCAPLUS

CN Octane, 1,1'-oxybis- (9CI) (CA INDEX NAME)

 $Me^- (CH_2)_7 - O^- (CH_2)_7 - Me$

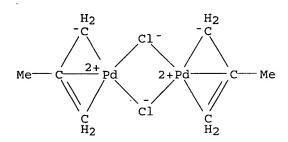
RN 7789-33-5 HCAPLUS

CN Iodine bromide (IBr) (6CI, 8CI, 9CI) (CA INDEX NAME)

Br-I

RN 12081-18-4 HCAPLUS

CN Palladium, di- μ -chlorobis[(1,2,3- η)-2-methyl-2-propenyl]di- (9CI) (CA INDEX NAME)



RN 14781-45-4 HCAPLUS

CN Copper, bis(1,1,1,5,5,5-hexafluoro-2,4-pentanedionato- κ O, κ O')-, (SP-4-1)- (9CI) (CA INDEX NAME)